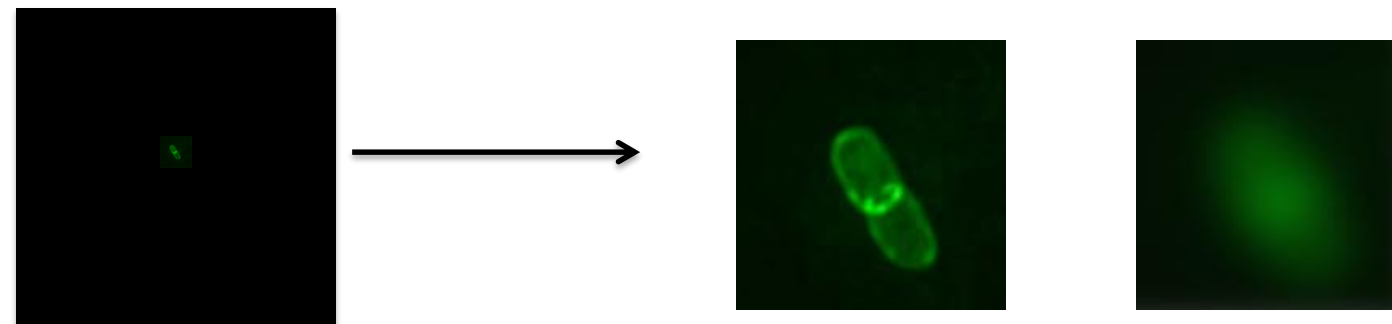


*Week 6-7*

# Fluorescence and FRET

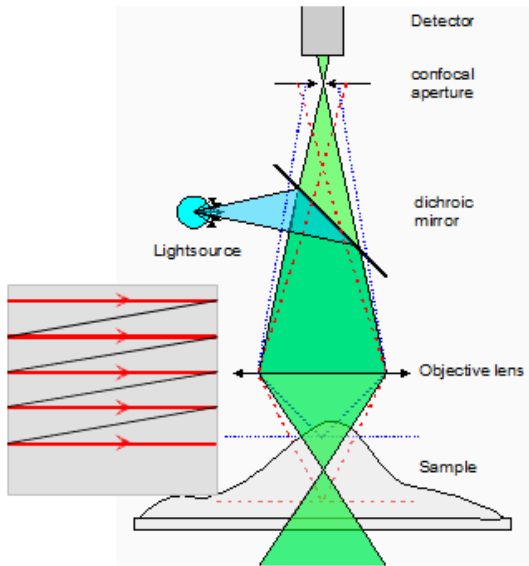
--



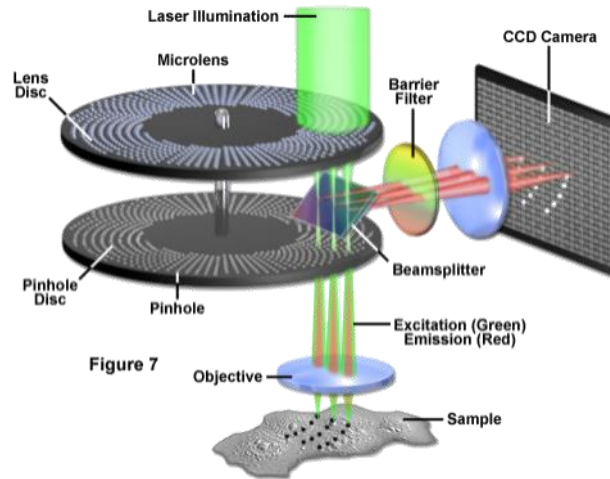
# Diffraction limited high resolution microscope Methods

1. Laser scanning confocal
2. Spinning disc confocal
3. Multi-photon confocal microscopy

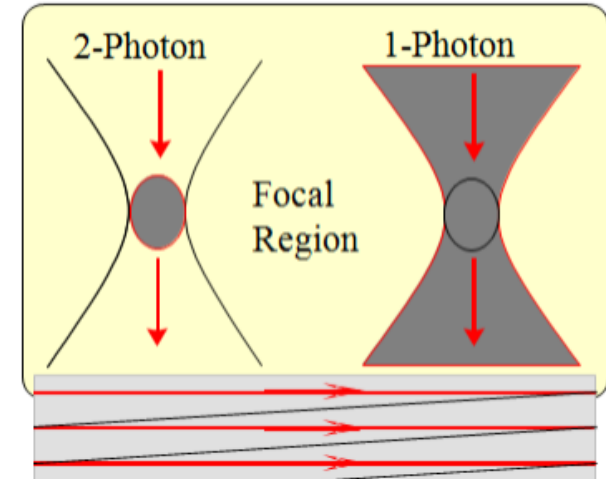
## Point Scanning Confocal



## Spinning Disk Confocal

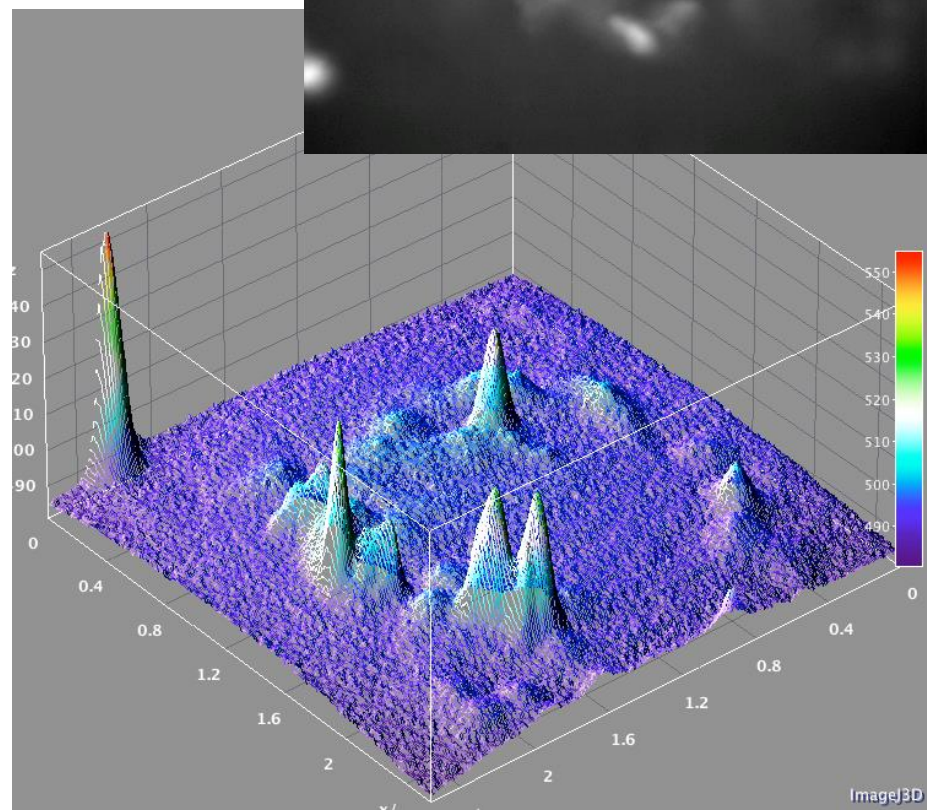
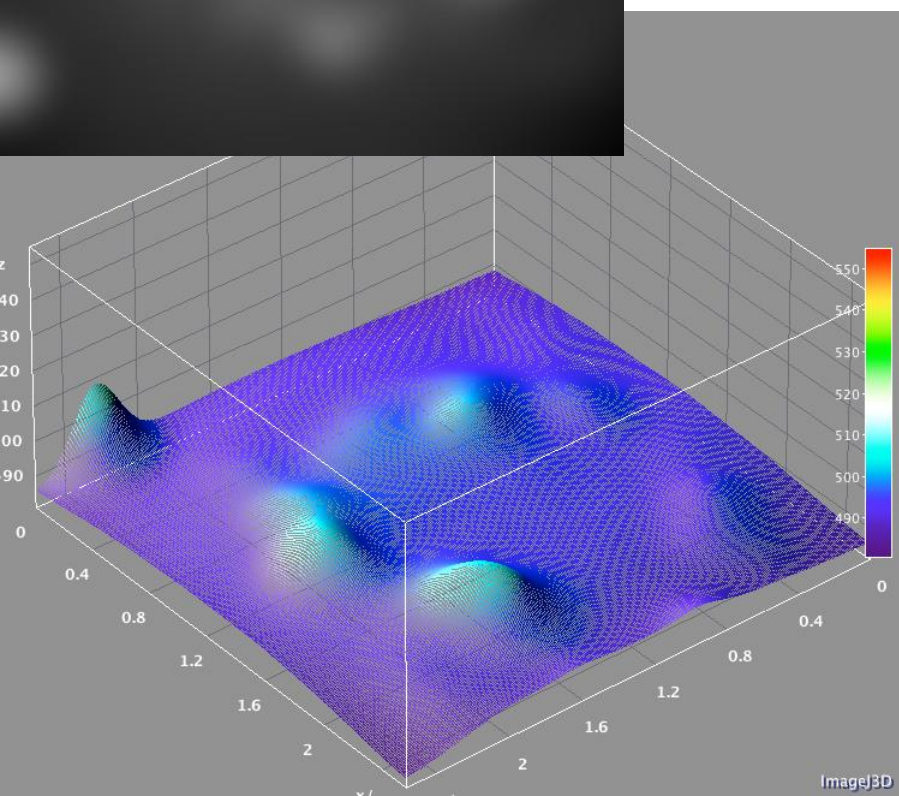
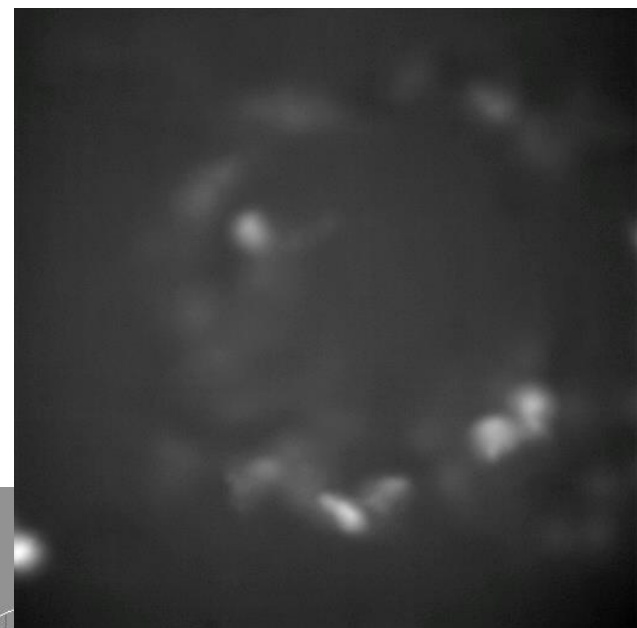
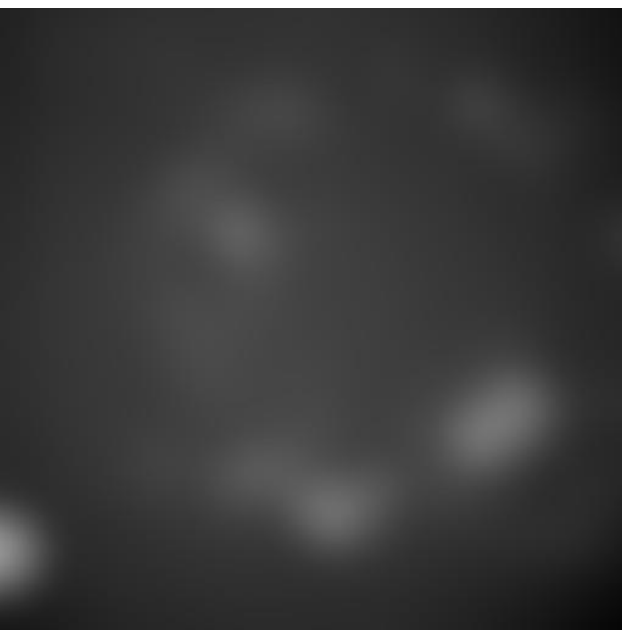


## Multiphoton



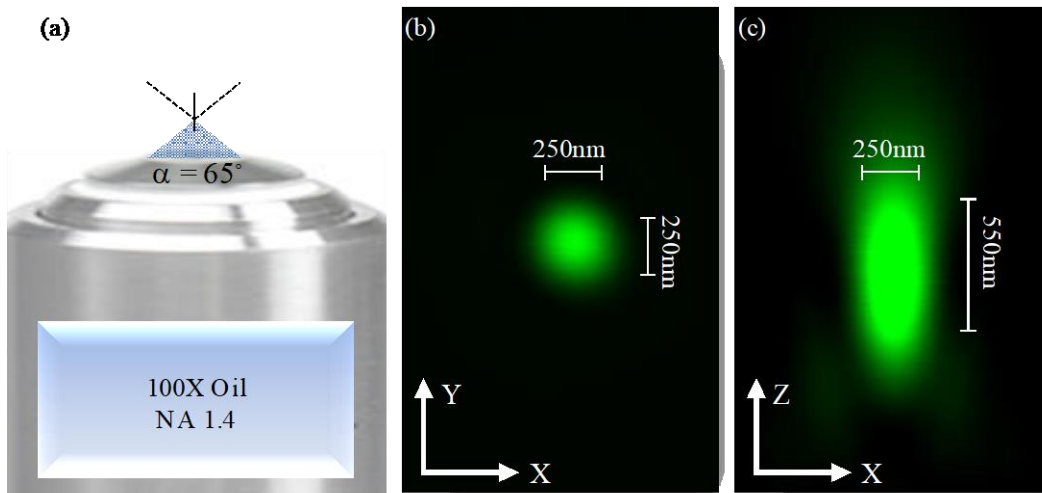
# Point Spread Function

GFP expressing cells

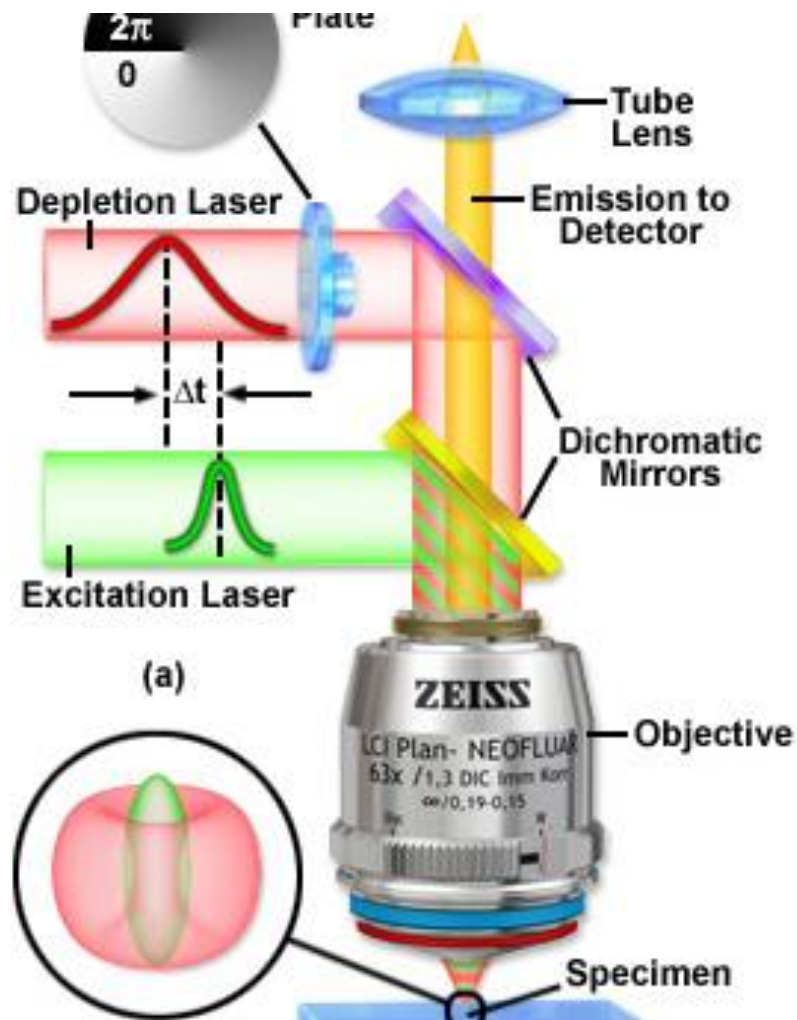


# Super Resolution Microscopy

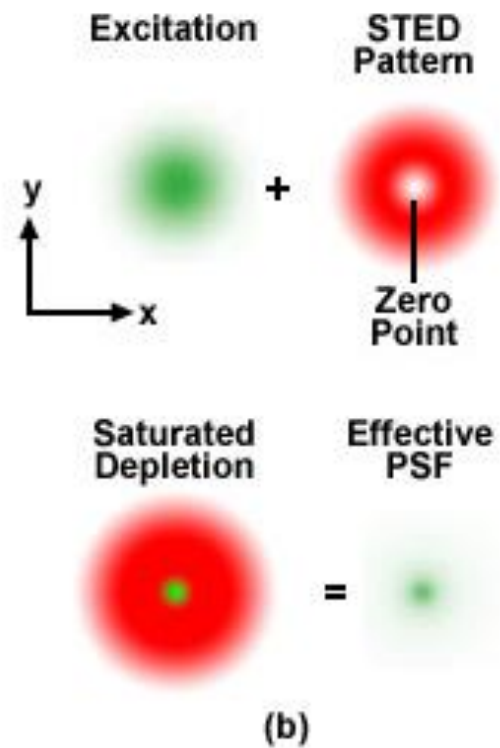
- STORM : Stochastic Optical Reconstruction Microscope
- PALM : Photo activated Localization Microscope
- STED: Stimulated Emission Depletion



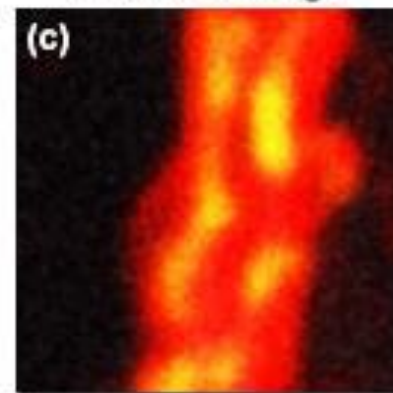
Can we reduce the size of the point spread function?  
What other solutions are possible?



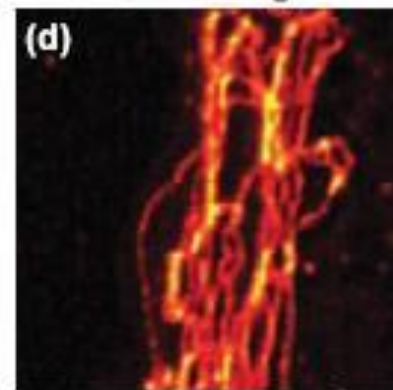
### STED Microscope Point-Spread Functions



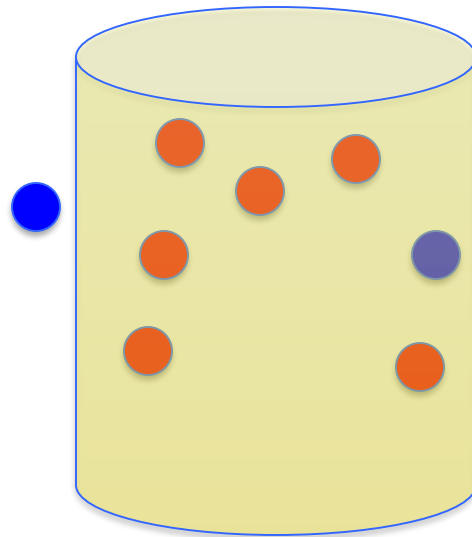
### Widefield Image



### STED Image



Many dye molecules are excited that reduce any good resolution. As the light of the different dye molecules can not be distinguished.



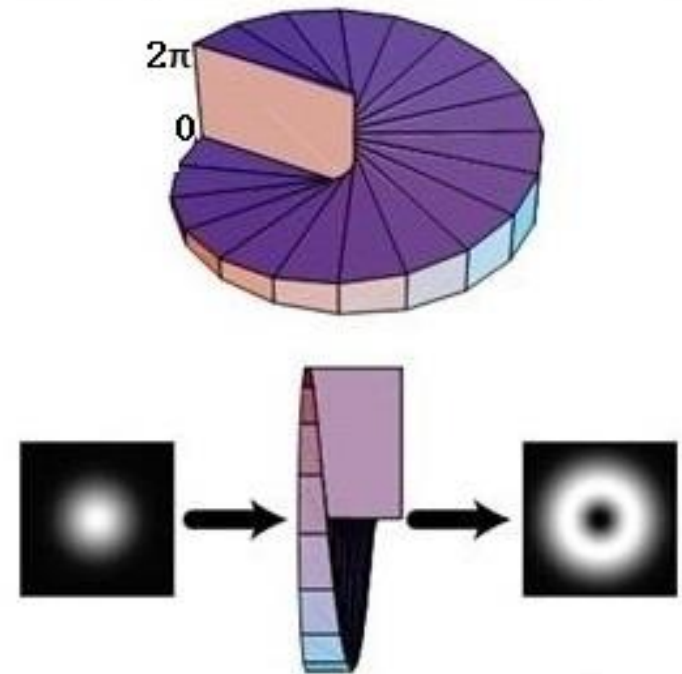
Many molecules are excited by a laser spot.



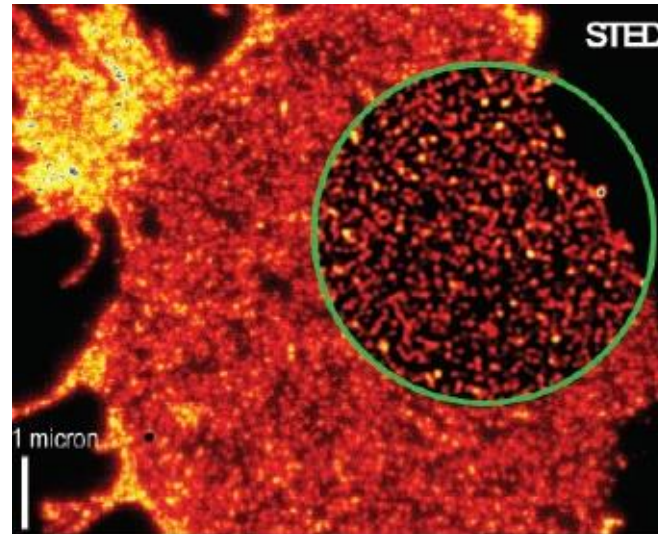
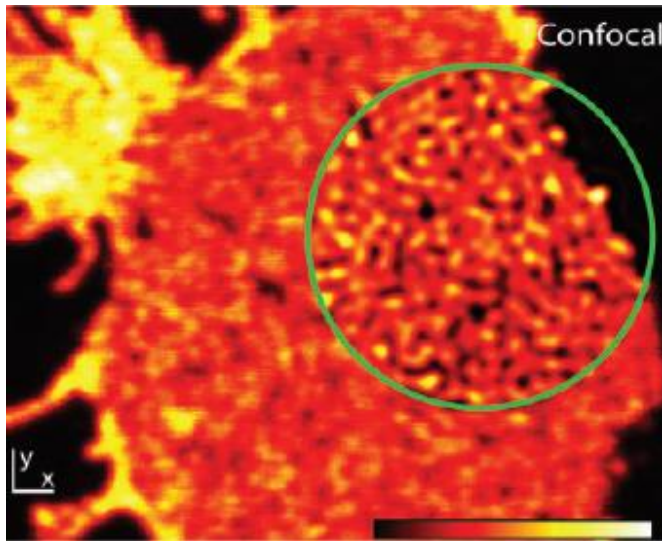
# How to make a hole at the center of the beam?

- Optical vortex method can be used to make a twisted light with a center in the hole. Phase plate can be used to make optical vortex.
- Optical vortex has a zero of an optical field.

Courtesy of Courtial and O'Holleran, 2007



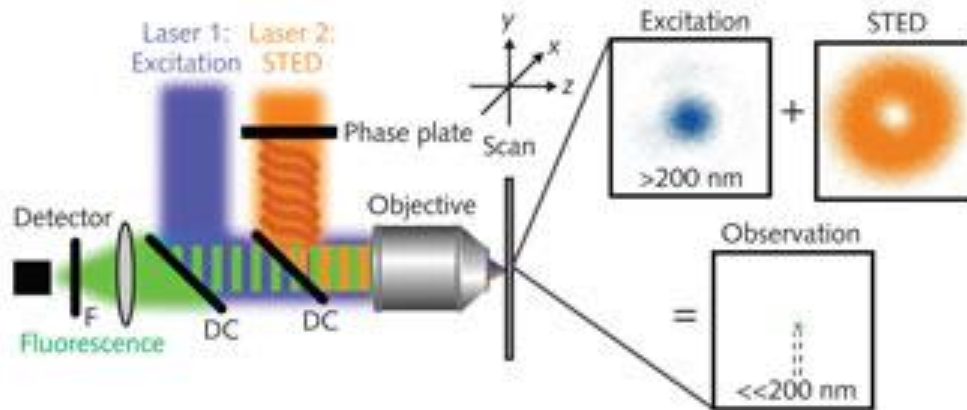
# STED vs Confocal Microscope



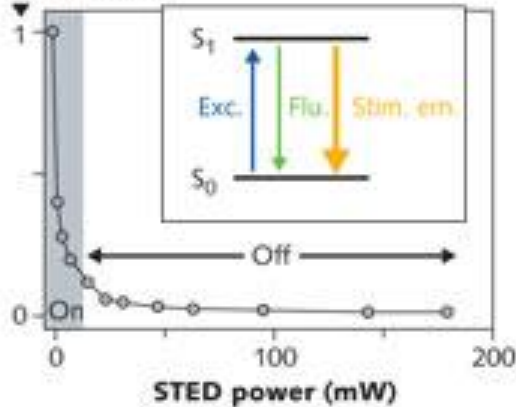


# STED Setup

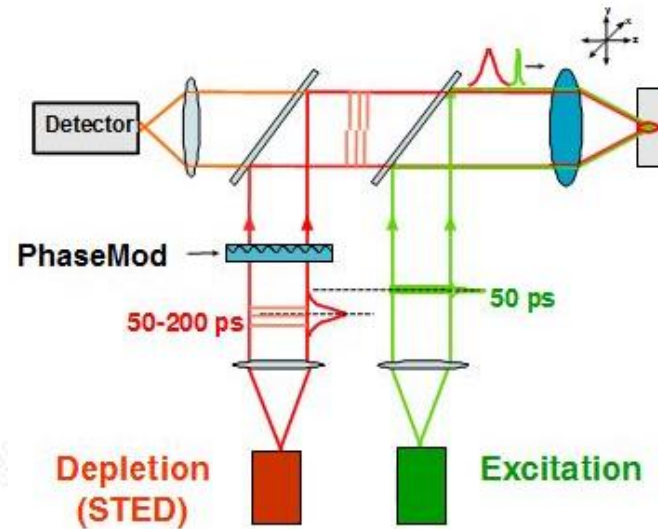
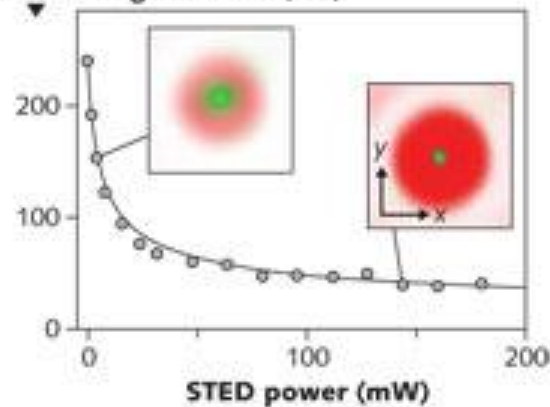
a)



b) Fluorescence



c) Observing diameter (nm)

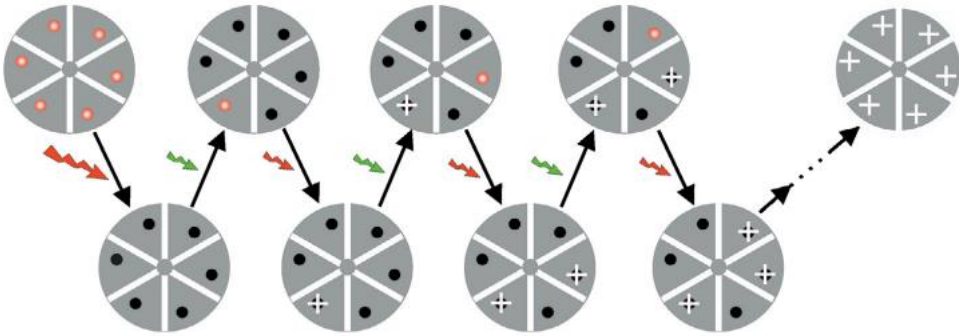


The power dependence delivers subdiffraction size observation volumes on the spot. The volume in which fluorescence emission is allowed (green, insets) but decreases with increasing STED laser power.

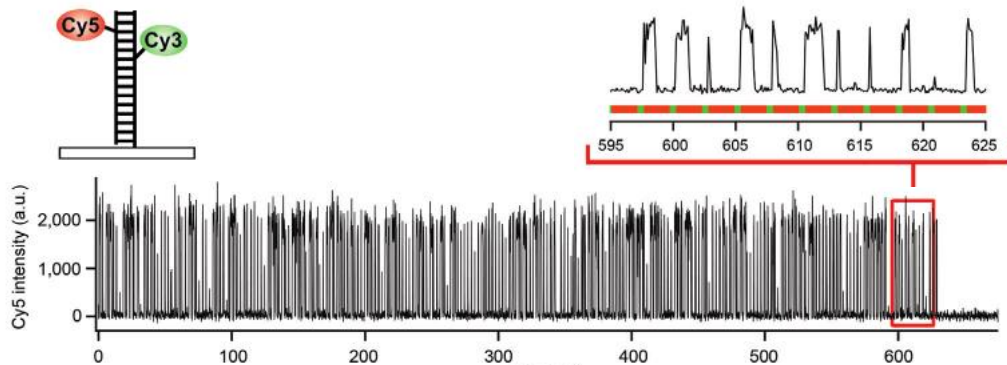


# Each point are imaged stochastically

a



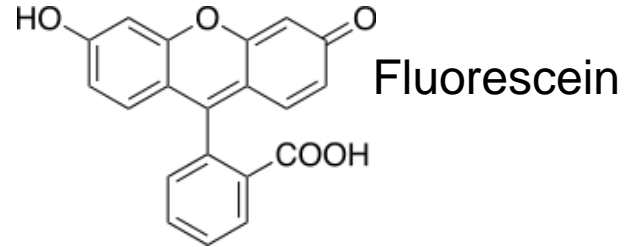
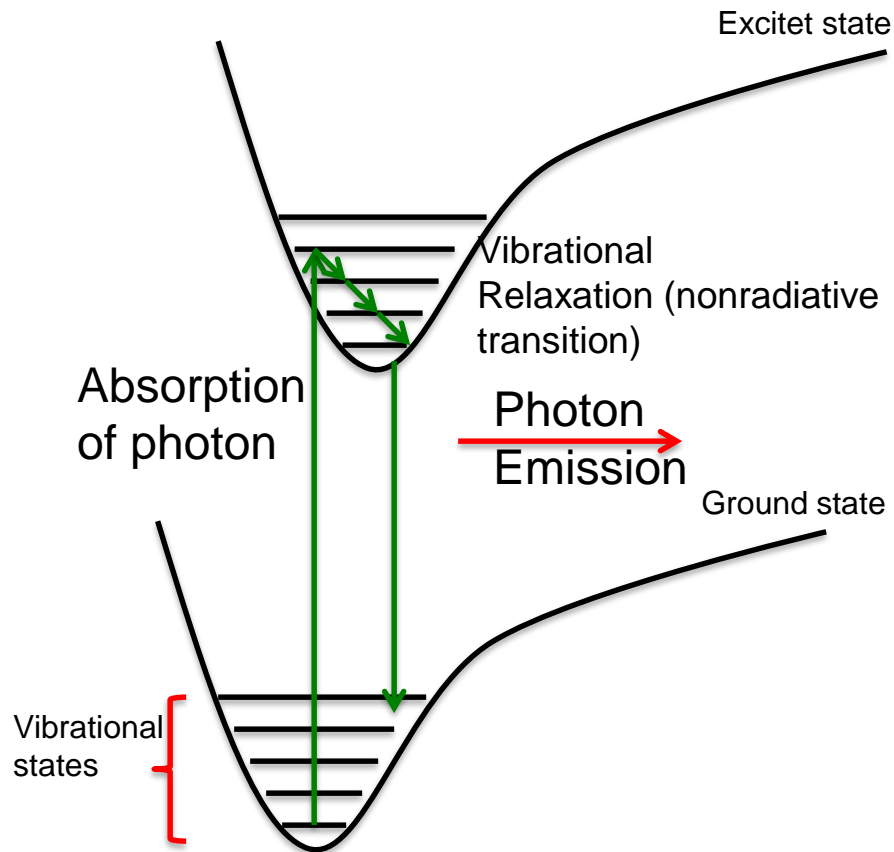
b



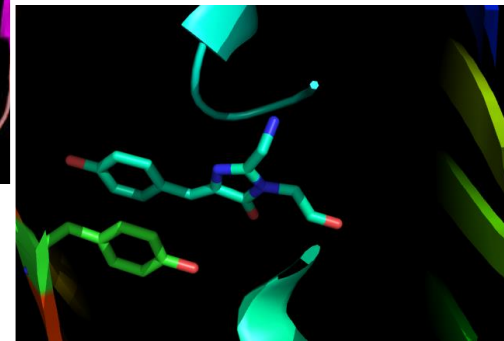
# **Fluorescence Resonance Energy Transfer (FRET) and its applications**

# Fluorescence

Electronic transition during fluorescence



Green  
Fluorescent  
Protein



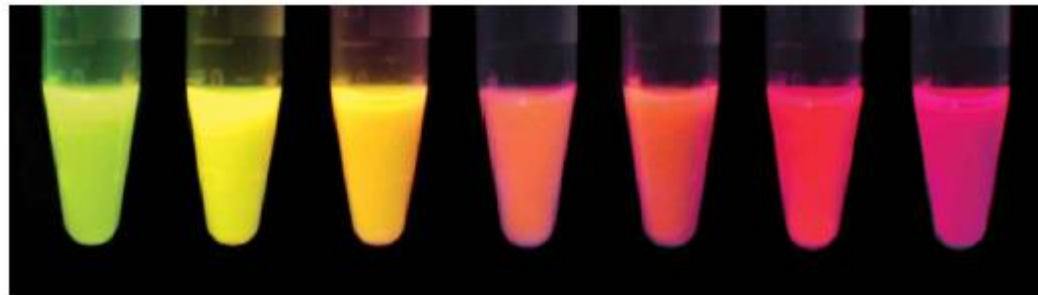
# Different “GFPs”

Absorption



d

Fluorescence



mHoneydew, mBanana, mOrange, tdTomato, mTangerine,  
mStrawberry, mCherry



# Förster resonance energy transfer (FRET)

RADIATION RESEARCH SUPPLEMENT 2, 326-339 (1960)

## Transfer Mechanisms of Electronic Excitation Energy

TH. FÖRSTER

*Laboratorium für physikalische Chemie der Technischen Hochschule, Stuttgart, Germany*

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II. Possible Transfer Mechanisms.....	328
III. General Principles of Resonance Transfer.....	328
IV. Theory of Slow Resonance Transfer.....	330
V. Resonance Transfer in Proteins.....	335
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### I. EXPERIMENTS ON ENERGY TRANSFER IN BIOLOGICAL SYSTEMS

For many years, special mechanisms of *energy transfer* have been discussed in relation to biological systems. It was thought necessary to assume that energy liberated at one molecule or at one distinct unit of a complex protein can be effective at another molecule or at another unit of the same protein molecule well-separated from the first one.

In order to account for such transfer, quite different mechanisms have been proposed, such as a successive shift of protons over internal hydrogen bridges (1) connecting different amino acids in a protein, or the free motion of electrons supposed to be common to the entire protein (2). These mechanisms are not considered in detail in the following, as this has been done well enough by other authors [cf. Bücher (3) or Vladimirov and Konev (4)]. Instead, discussed below are those experiments which, in my opinion, unequivocally show the existence of energy transfer in biological systems. From these experiments, and from the general properties of the systems involved, an attempt is made here to draw certain conclusions on the possible transfer mechanisms and to evaluate these mechanisms in detail.





Energy transfer has been discussed in relation to oxidative metabolism. In this instance, however, it seems from recent investigations, that the energy resides in energy-rich molecules of low molecular weight which are free to move from one enzyme to another. Energy transfer has been considered further in connection with the

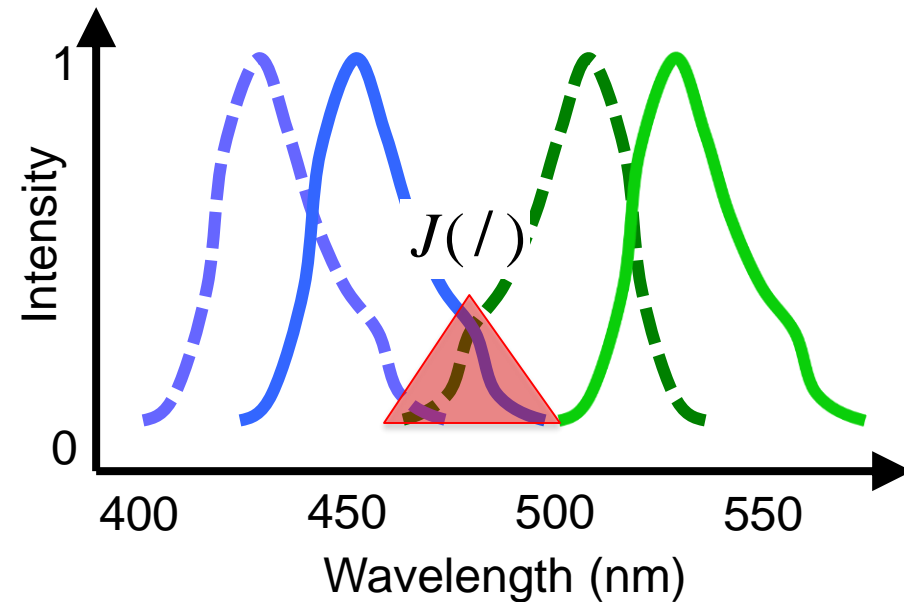
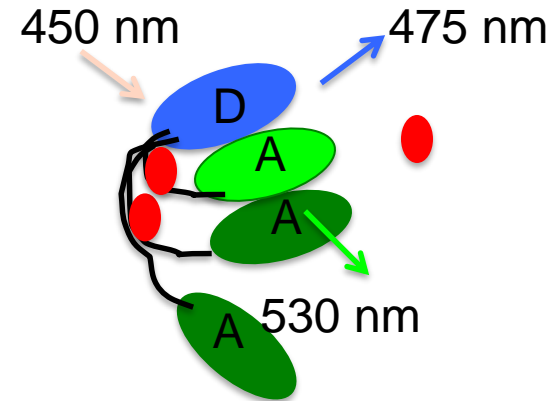


Theodor Förster  
1910 - 1974

Theodor Förster

# Fluorescence Resonance Energy Transfer (FRET)

-  Excitation of donor fluorescent probe
-  Emission of donor fluorescent probe
-  Excitation of acceptor fluorescent probe
-  Emission of acceptor fluorescent probe

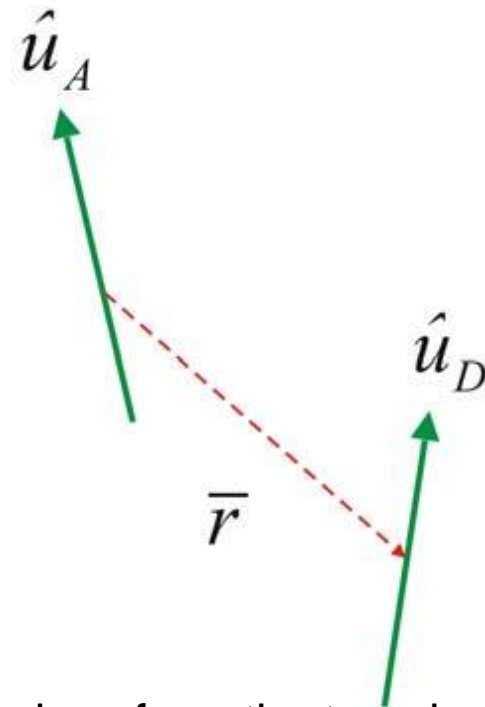
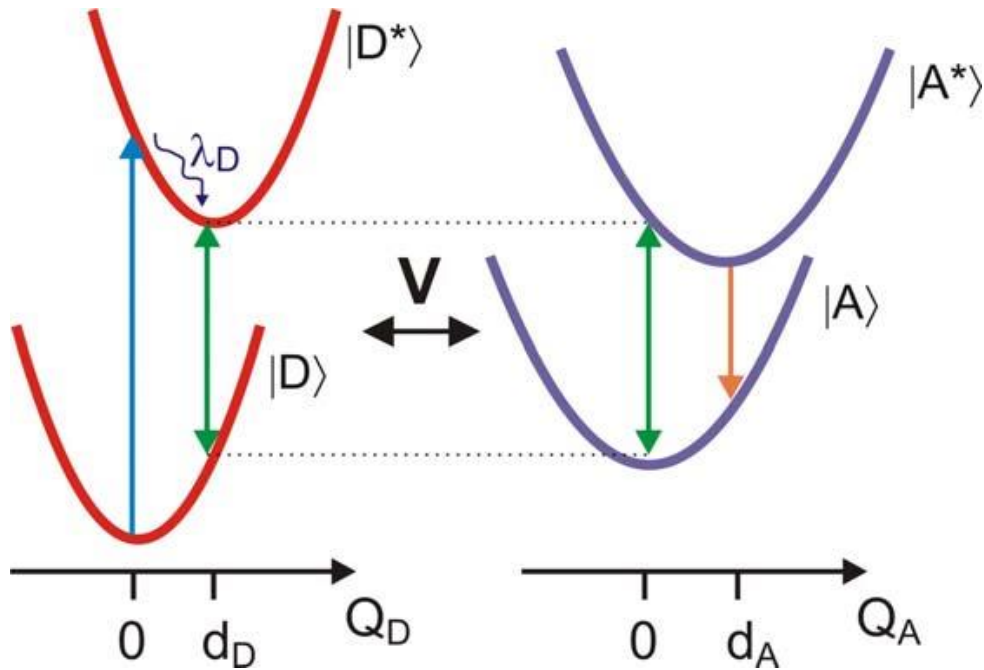


$$E = \frac{1}{1 + \left(\frac{r}{R_0}\right)^6}$$

$$R_0 = (\kappa^2 \cdot J(\lambda) \cdot n^{-4} \cdot Q)^{1/6}$$

$J(l)$  = Degree of Spectral Overlap

# Nonradiative energy transfer between two particles



FRET arises from the transient dipole interaction of donor and acceptor.

FRET is a nonradiative energy transfer from donor to acceptor.

# Quantum yield of fluorophore

The fluorescence quantum yield ( $\Phi_F$ ) is the ratio of photons absorbed to photons emitted through fluorescence.

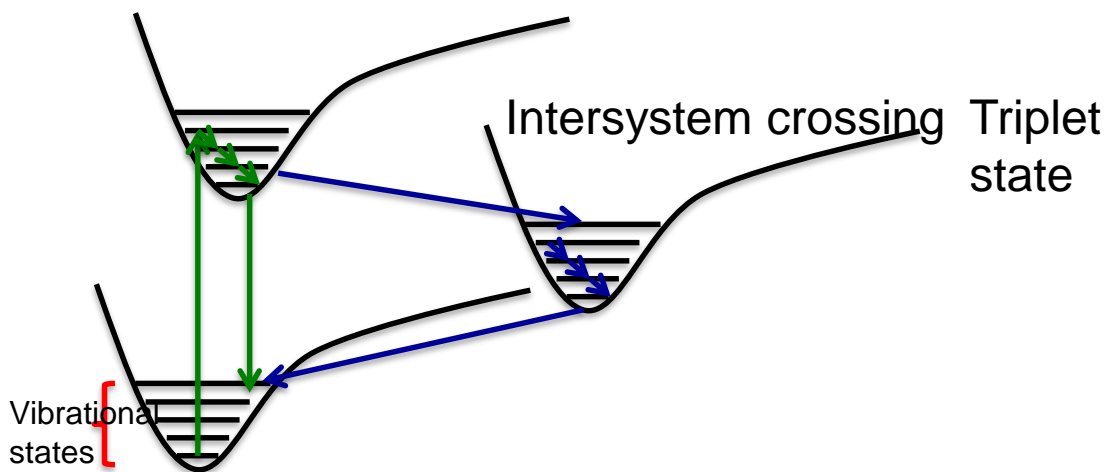
$$F = \frac{\text{number of photons emitted}}{\text{number of photons absorbed}}$$

$$F = \frac{k_f}{k_f + k_{nr} + k_{isc}}$$

$k_f$  = rate of fluorescence

$k_i$  = rate of nonradiative decay

$k_{isc}$  = rate of intersystem crossing

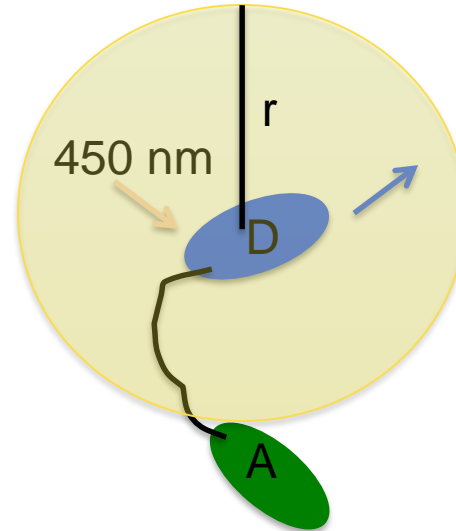


# Distance and spectral overlap effect

## FRET

The distance which can be probed by FRET is limited typically 1-10nm.

$$E = \frac{1}{1 + \left( \frac{r}{R_0} \right)^6}$$

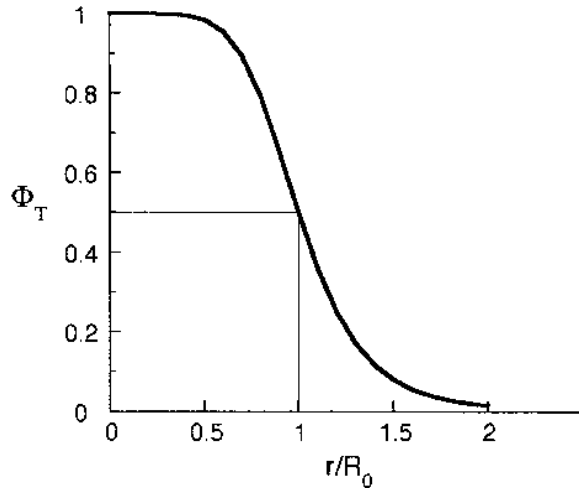


Due to the  $1/R^6$  distance dependence, distances in the range 0.5-  
 $1.5 R_0$

Efficiency in the range 0.98-0.10, are suitable for FRET measurements.

# Distance effect between donor and acceptor

Transfer efficiency is 50% when the donor-acceptor distance is equal to the Förster critical radius.



$$r = \left( \frac{1}{\Phi_T} - 1 \right)^{1/6} R_0$$

Tab. 8.4. Examples of Förster critical radii<sup>a)</sup>

<i>Donor</i>	<i>Acceptor</i>	<i>R<sub>0</sub></i> (nm) <sup>b)</sup>
Naphthalene	Dansyl	2.2
Anthracene	Perylene	3.1
Pyrene	Perylene	3.6
Phenanthrene	Rhodamine B	4.7
Fluorescein	Tetramethylrhodamine	5.5
Fluorescein-5-isothiocyanate	Eosin maleimide	6.0
Rhodamine 6G	Malachite Green	6.1
Europium (III) complex	Cy5 (carboxymethylindocyanine-N-hydroxysuccinimidyl ester)	7.6
Europium (III) complex	CdSe/ZnS nanocrystals (quantum dots)	8.4 – 9.6
Terbium (III) complex	CdSe/ZnS nanocrystals (quantum dots)	8.5 – 9.8
Tryptophan	Dansyl	2.1
Tryptophan	ANS	2.3
Tryptophan	Anthroyl	2.5
Tryptophan	Pyrene	2.8
Pyrene	Pyrene	1.0
2-Ethyl-naphthoate	2-Ethyl-naphthoate	1.4
Anthracene	Anthracene	2.2
Perylene	Perylene	3.8
Rhodamine 101	Rhodamine 101	5.8



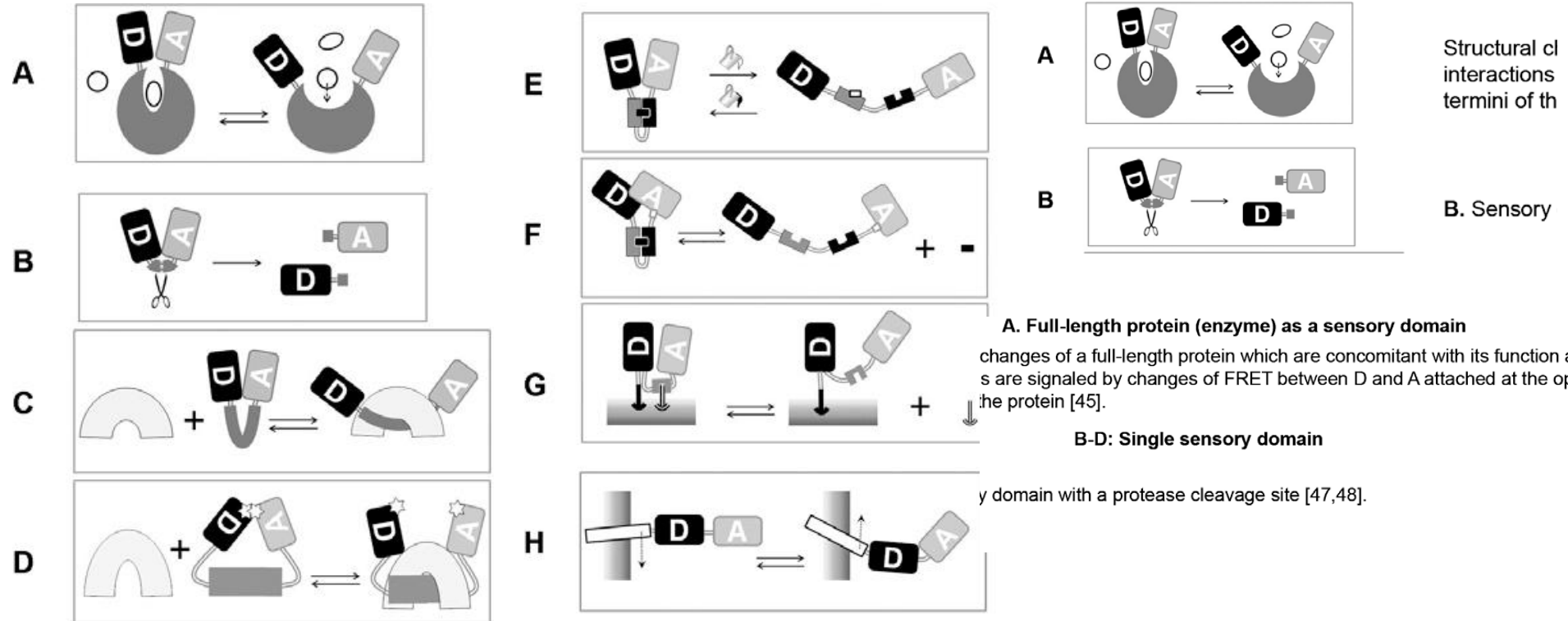
# FRET pair examples

Laser	Donor	Acceptor	Donor Ex Acceptor Em
Violet	Alexa Fluor 405	Alexa Fluor 430	405/541
Argon	Cy2	Cy3	488/566
Argon	Cy3	Cy5	488/666
Argon	FITC	TRITC	488/577
Argon	PE	APC	488/660
Argon	Alexa Fluor 488	Alexa Fluor 514	488/541
Argon	Alexa Fluor 488	Alexa Fluor 532	488/553
Argon	Alexa Fluor 488	Alexa Fluor 546	488/572
Argon	Alexa Fluor 488	Alexa Fluor 610	488/626
R-HeNe	Alexa Fluor647	Alexa Fluor 680	633/700
R-HeNe	Alexa Fluor647	Alexa Fluor 700	633/720
R-HeNe	Alexa Fluor647	Alexa Fluor 750	633/780

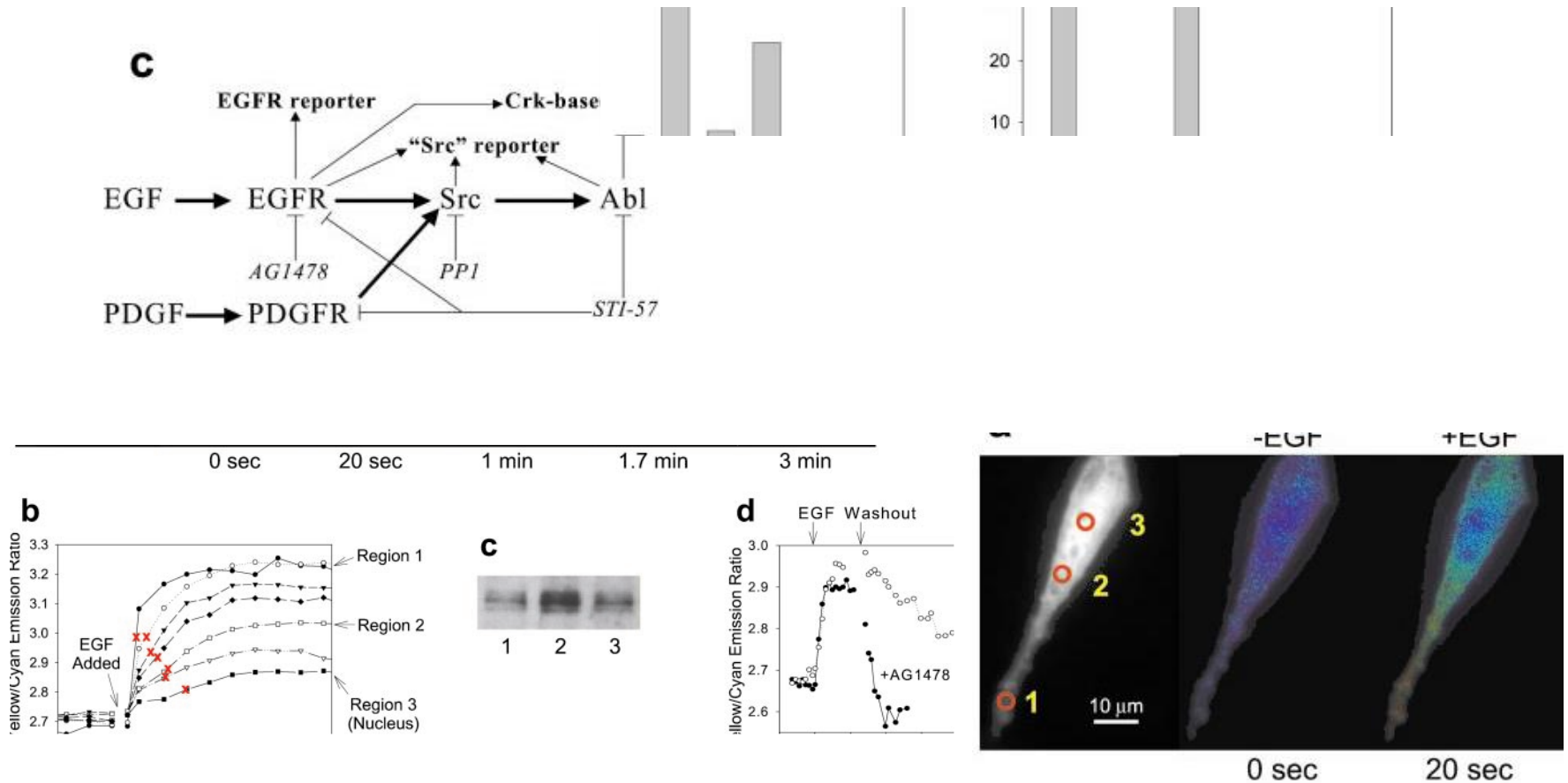
## FRET Pair Fluorescent Proteins

Laser	Donor	Acceptor	Donor Ex Acceptor Em
Violet	CFP	YFP	405/526
Violet	Cerulean FP	YFP	405/526
Argon	GFP	YFP	488/526
Argon	GFP	mRFP	488/579

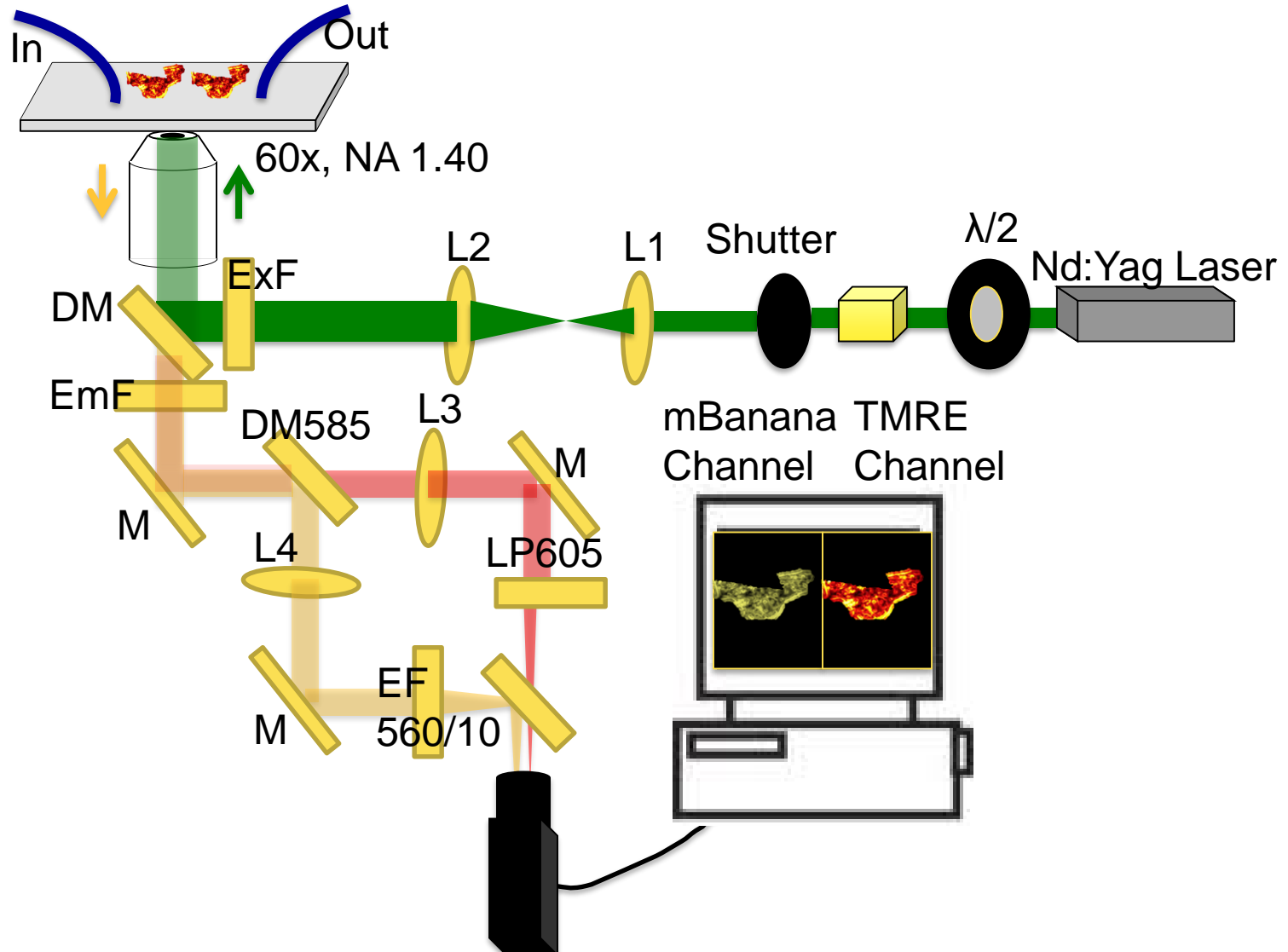
# FRET assays



# FRET Assay to monitor kinase activity in cells

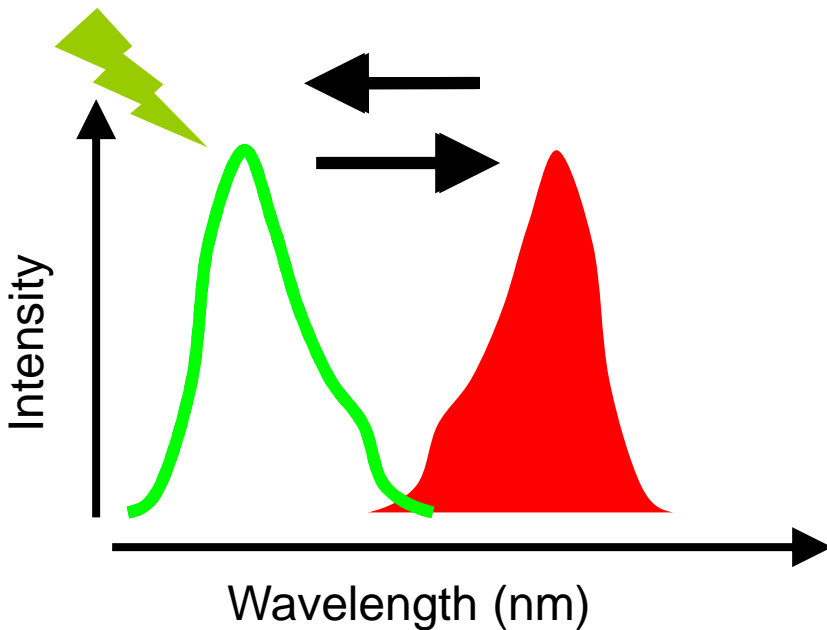


# Dual color imaging for FRET



# Theory of Photochromic FRET for sensing protein dynamics

- Ground state absorption of membrane protein
- - - Intermediate state absorption of membrane protein
- Emission of fluorescent probe (donor)

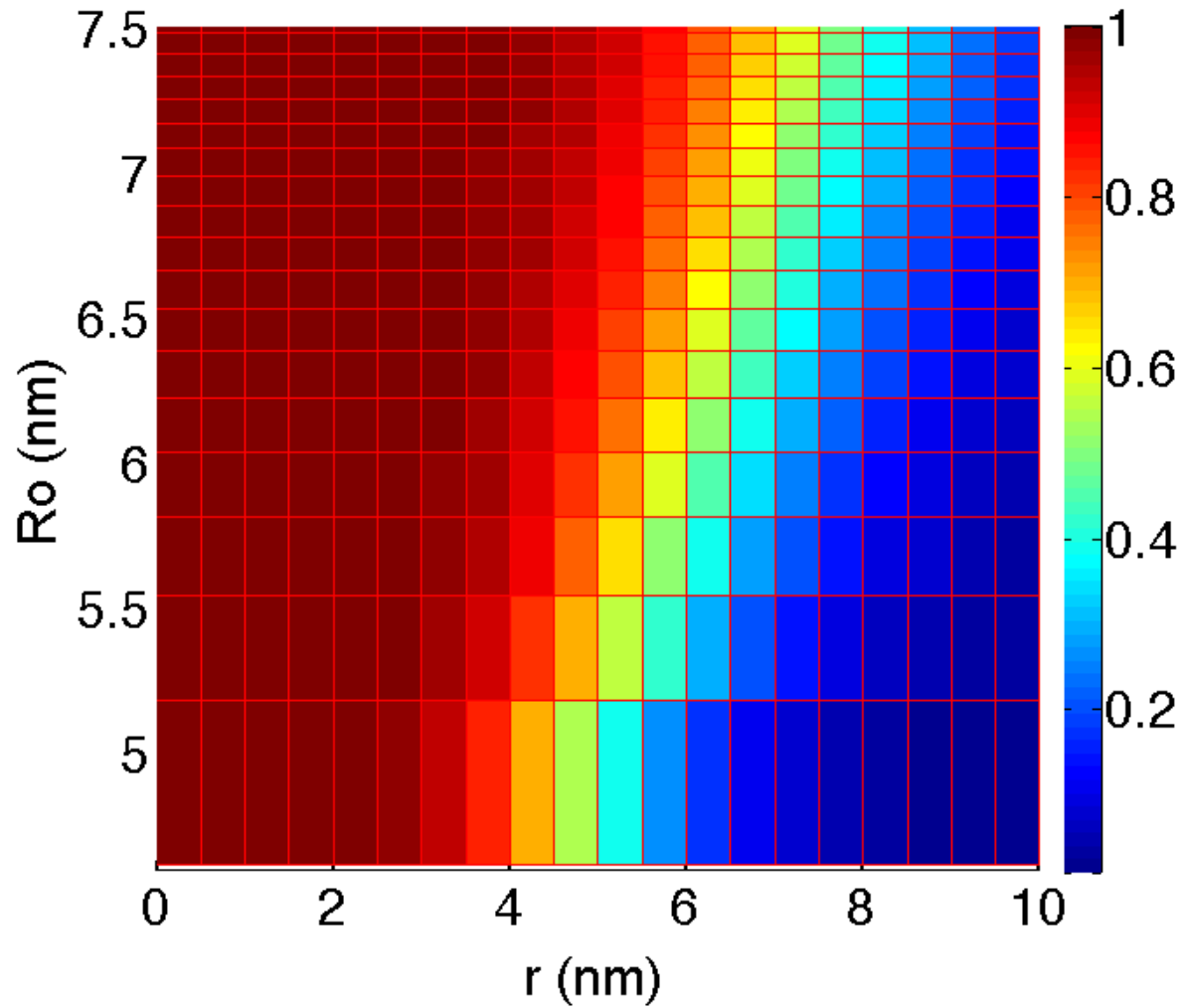


$$E = \frac{1}{1 + \left(\frac{r}{R_0}\right)^6}$$

$$R_0 = (\kappa^2 \cdot \boxed{J(\lambda)} \cdot n^{-4} \cdot Q)^{1/6}$$

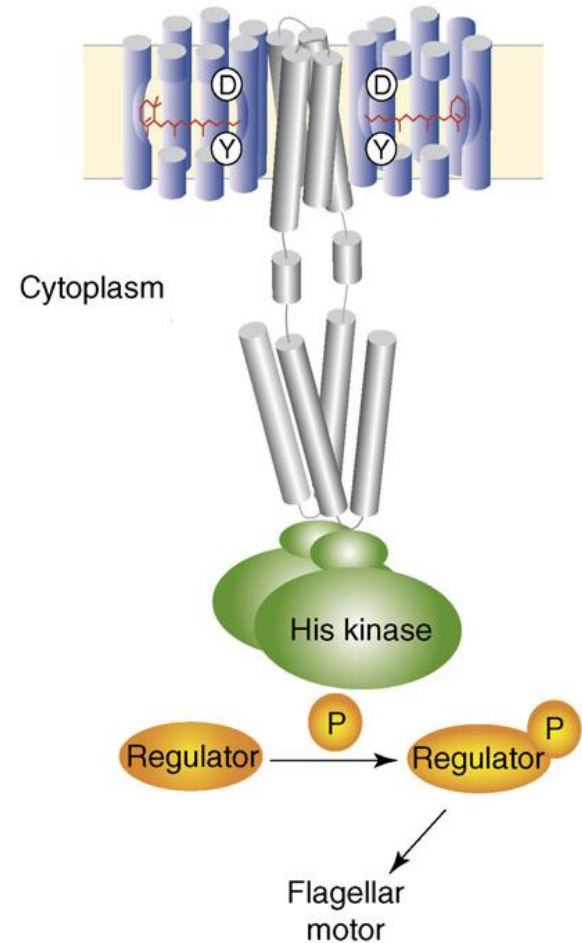
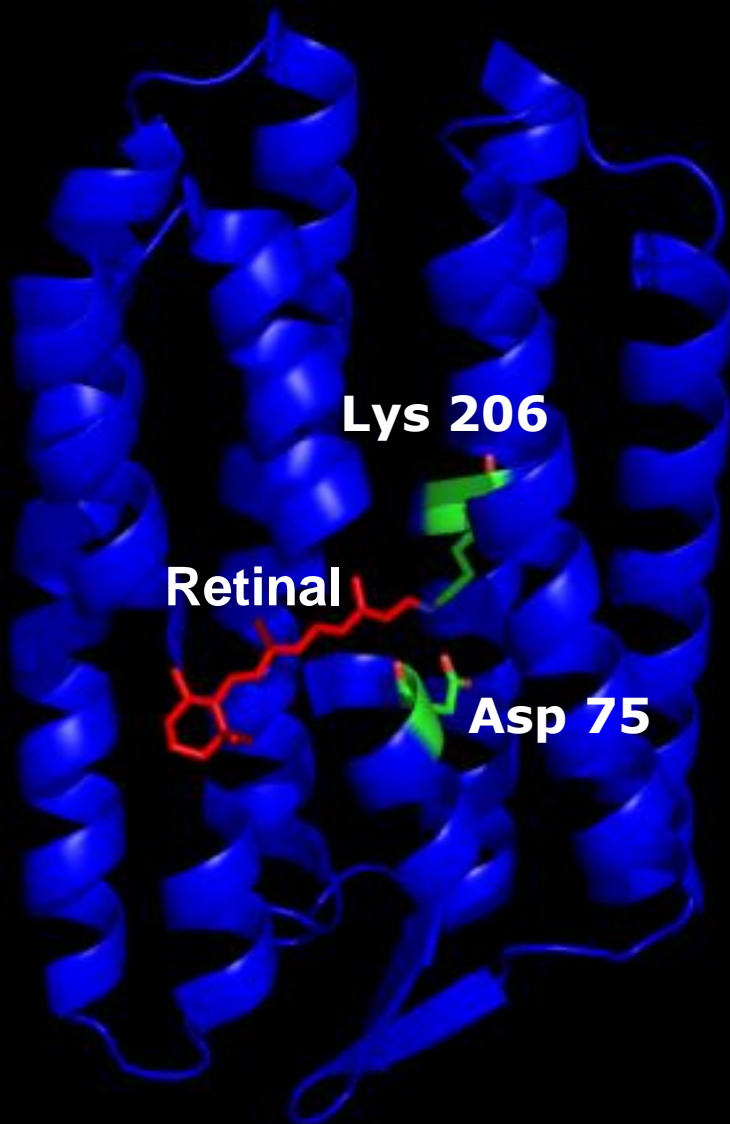
$J(\lambda)$  = Degree of Spectral Overlap

# pcFRET efficiency





# Crystal structure of sensory rhodopsin II

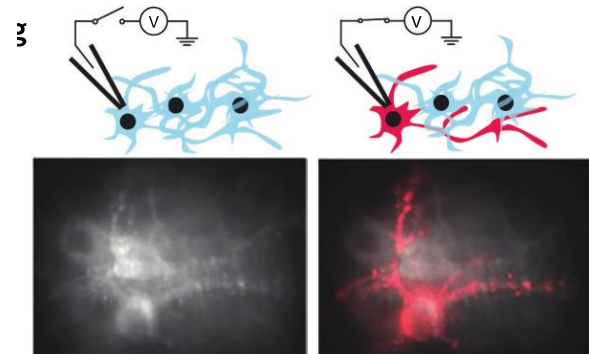
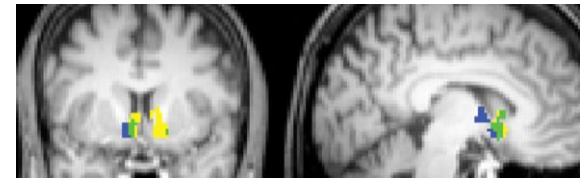


# Studying photosensitive membrane proteins

- Dynamic studies of photoactive membrane proteins are currently limited by absorption.
- Can we measure the photocycle of membrane protein in the cell? What does protein work in the cell membrane?
- How does the photocycle of a single membrane protein differ from the bulk?
- How different is the photocycle among single membrane proteins?

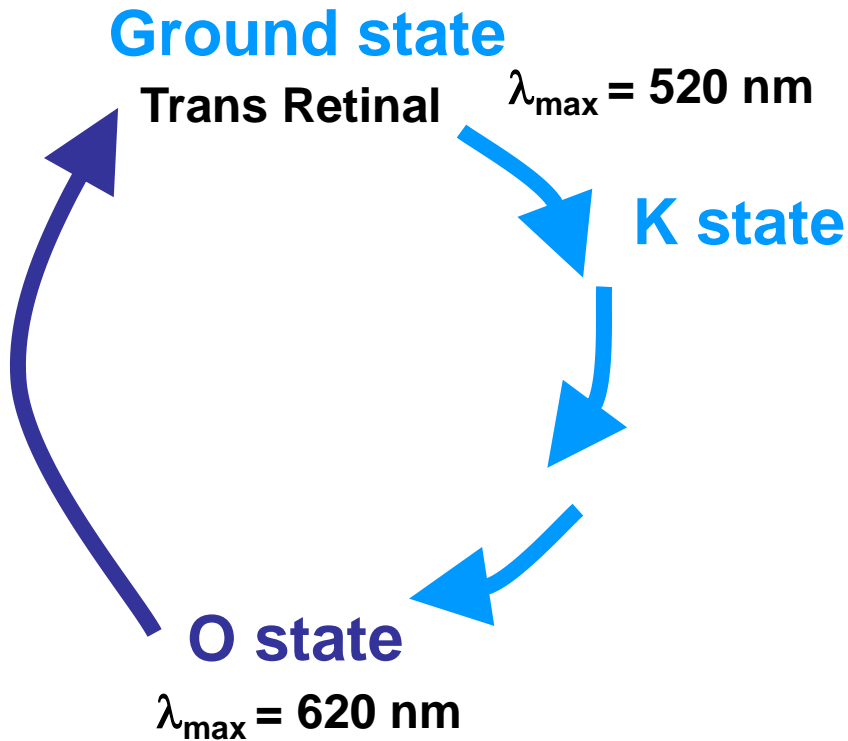
## Optogenetics

- Optical control of neuronal activity
- Optical probes for membrane environment

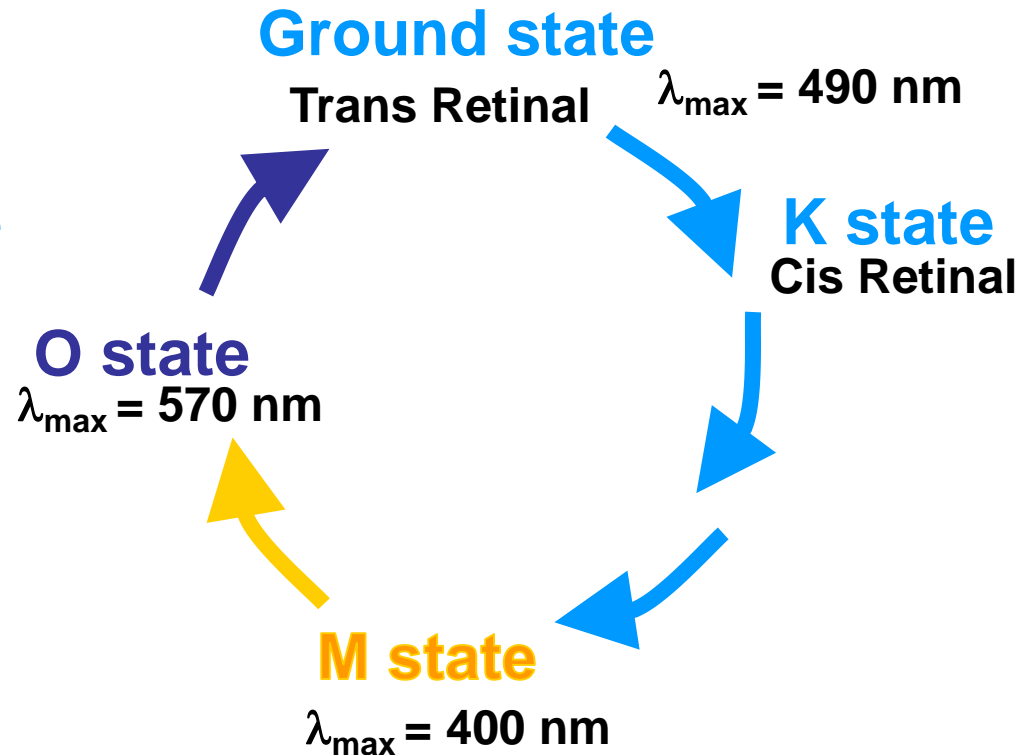


# Photocycle of photoactive membrane proteins

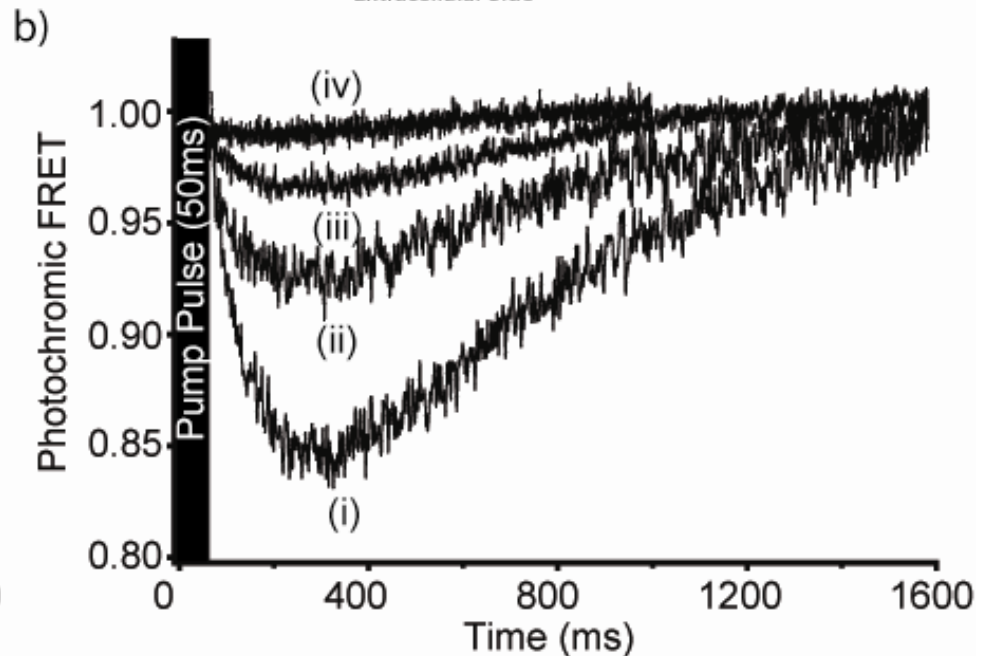
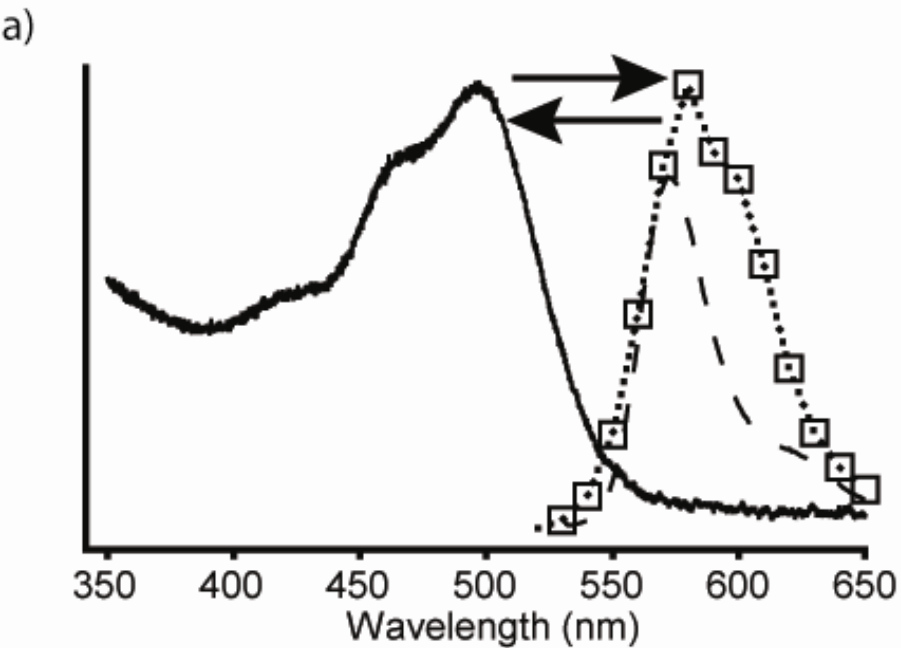
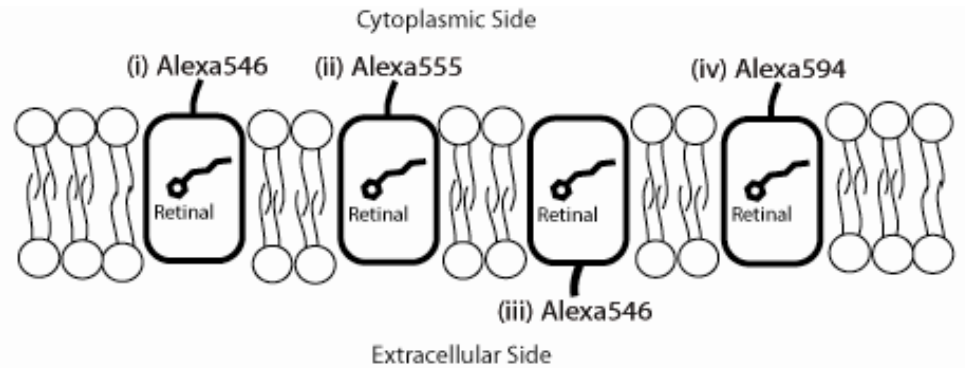
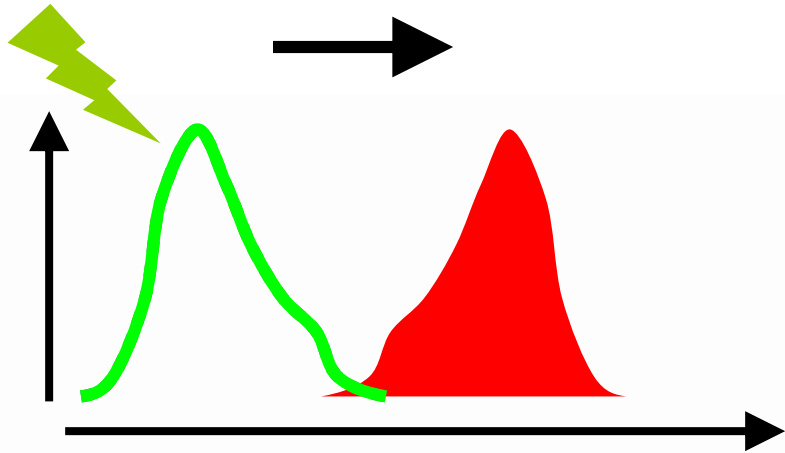
## Blue-Proteorhodopsin



## Sensory rhodopsin II

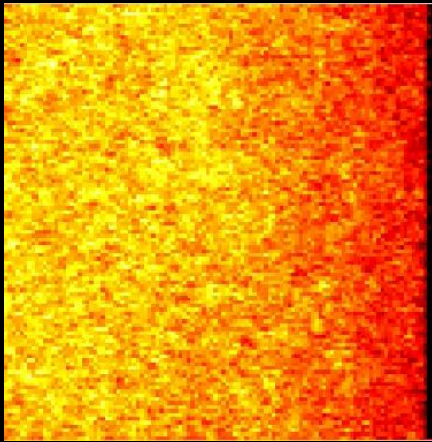


# Photochromic FRET for SRII

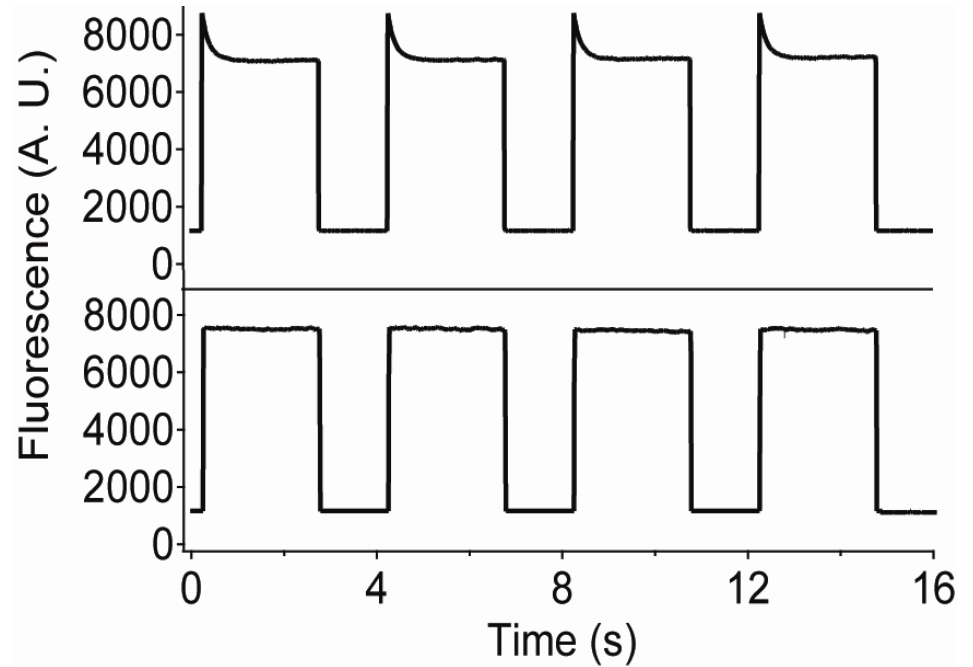
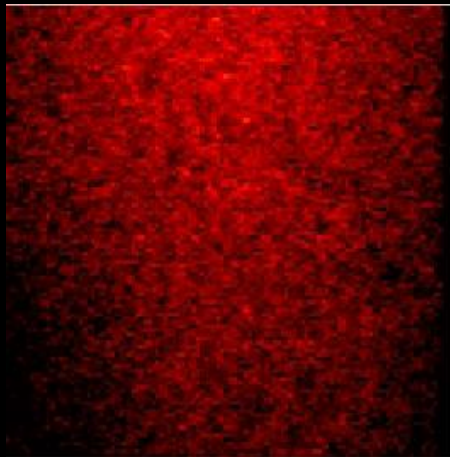


# O Intermediate state formation in SRII photocycle

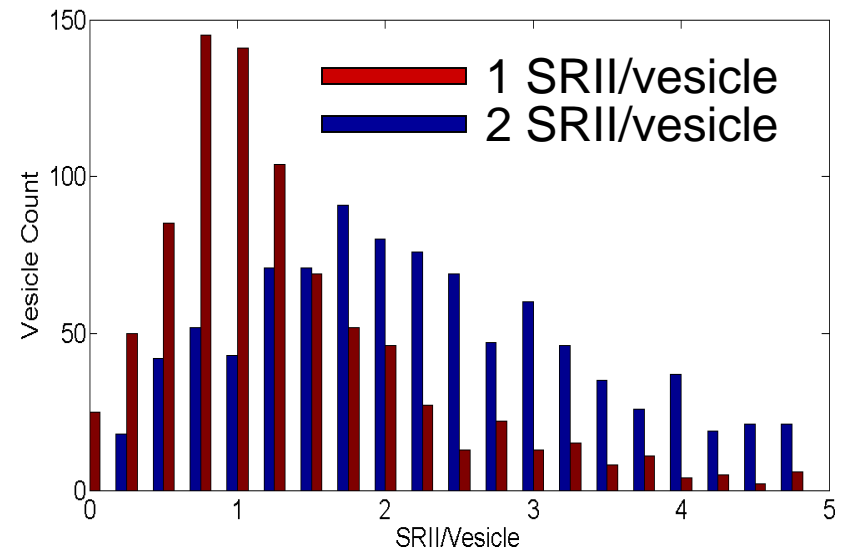
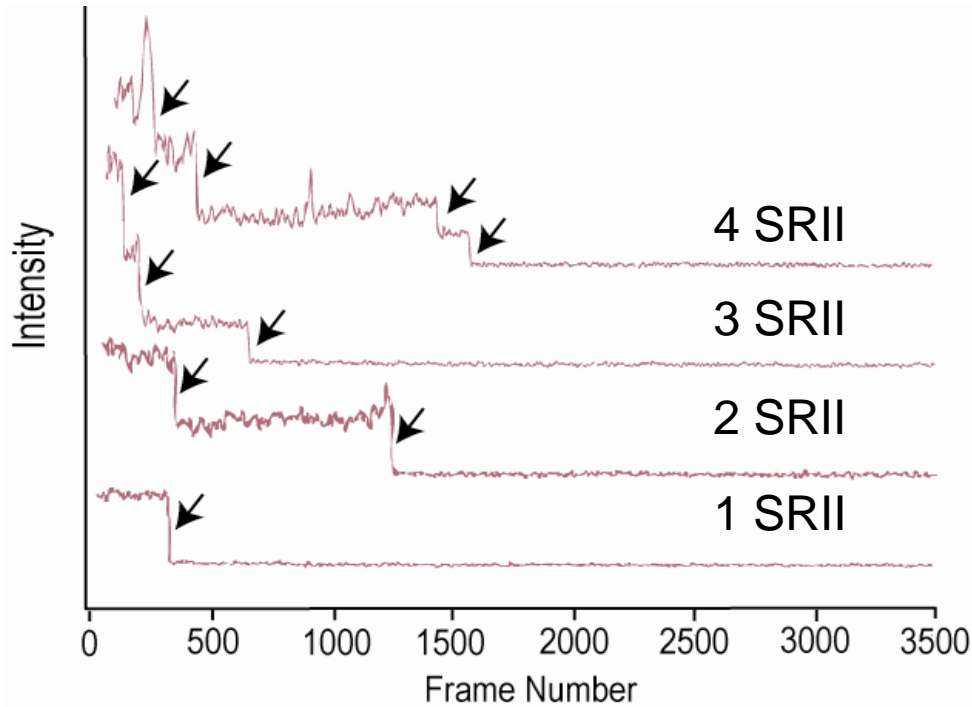
**SRIIAlexa546 (+Retinal)**



**Control = SRIIAlexa546 (-Retinal)**



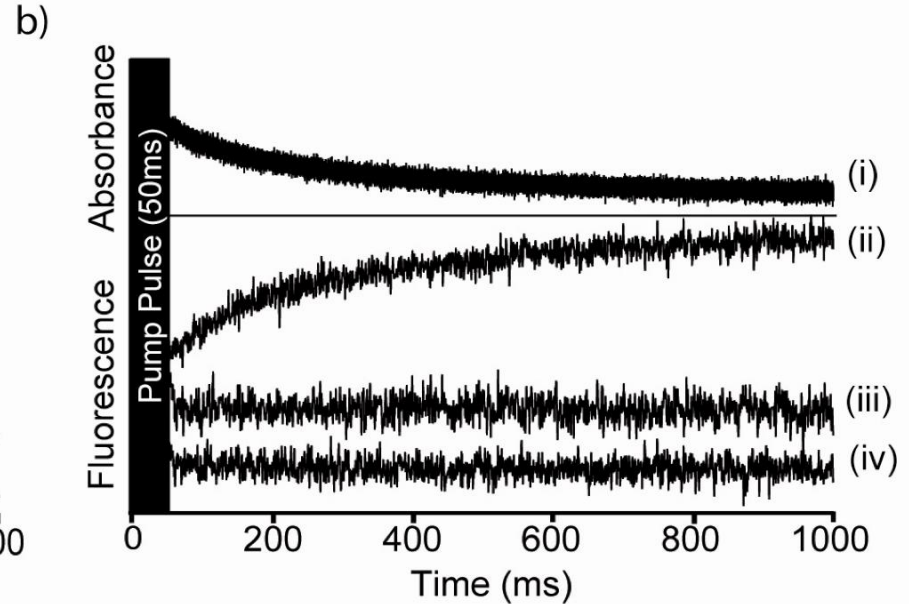
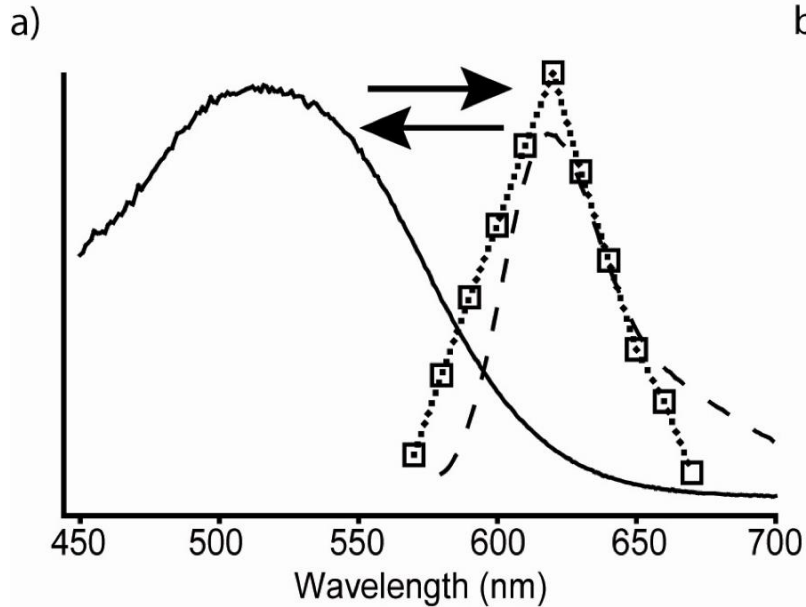
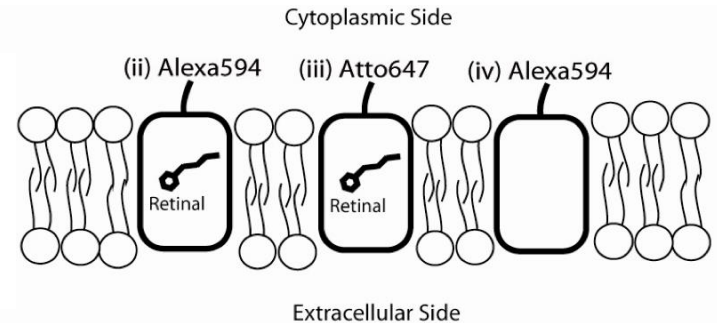
# Distribution of number of SRII in vesicles



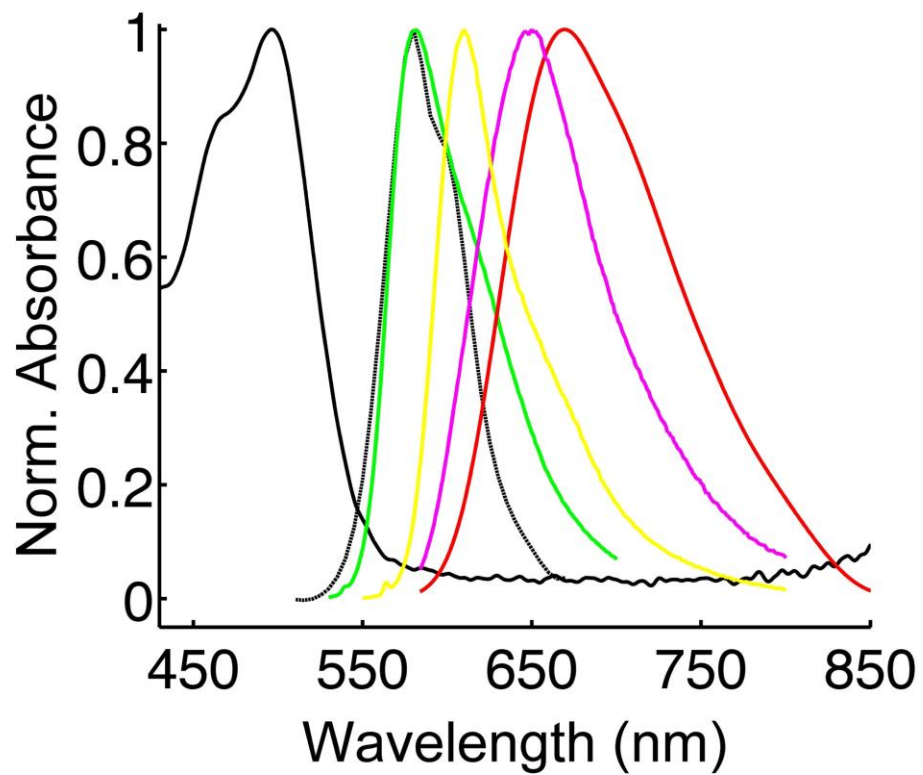


# Photochromic FRET for Blue-Proteorhodopsin

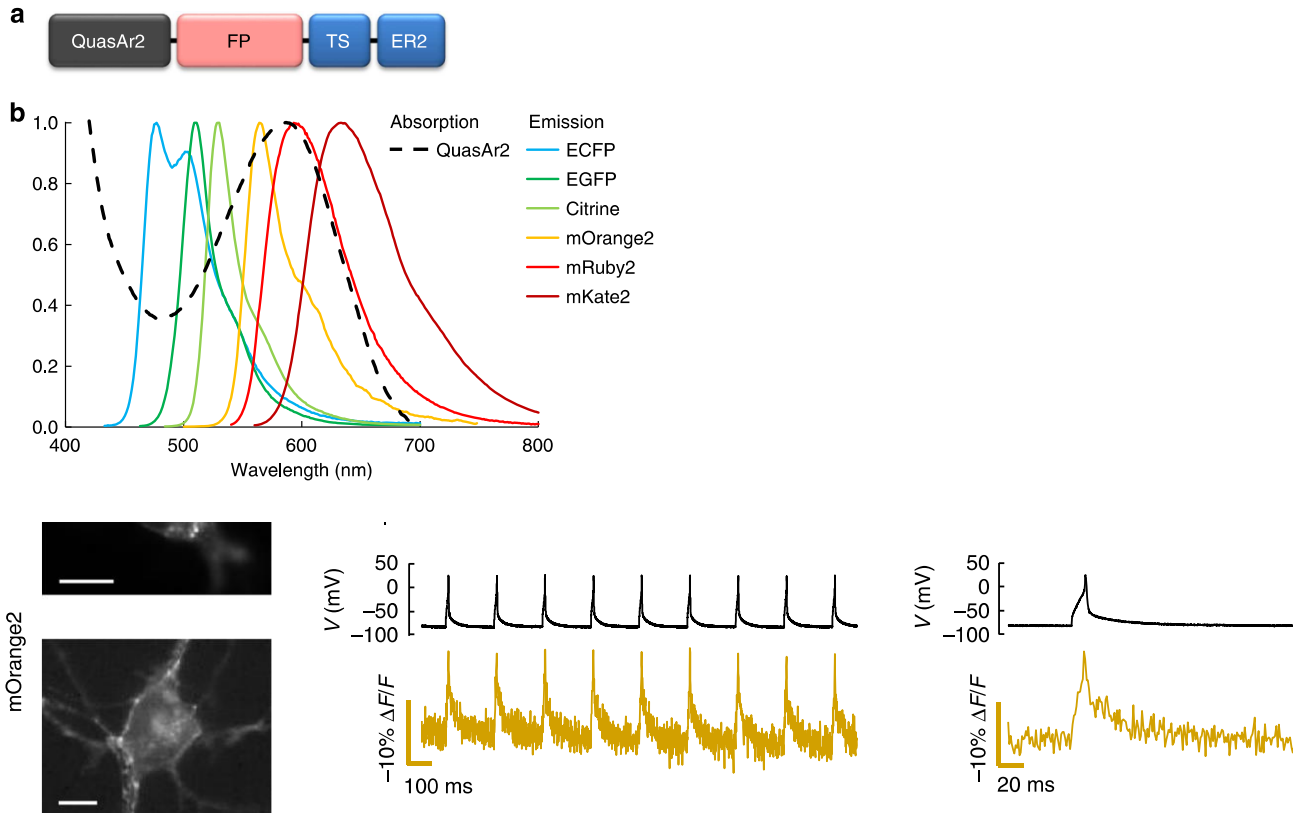
— Ground state absorption  
..... Intermediate state absorption  
----- Emission of Alexa 594



(A)



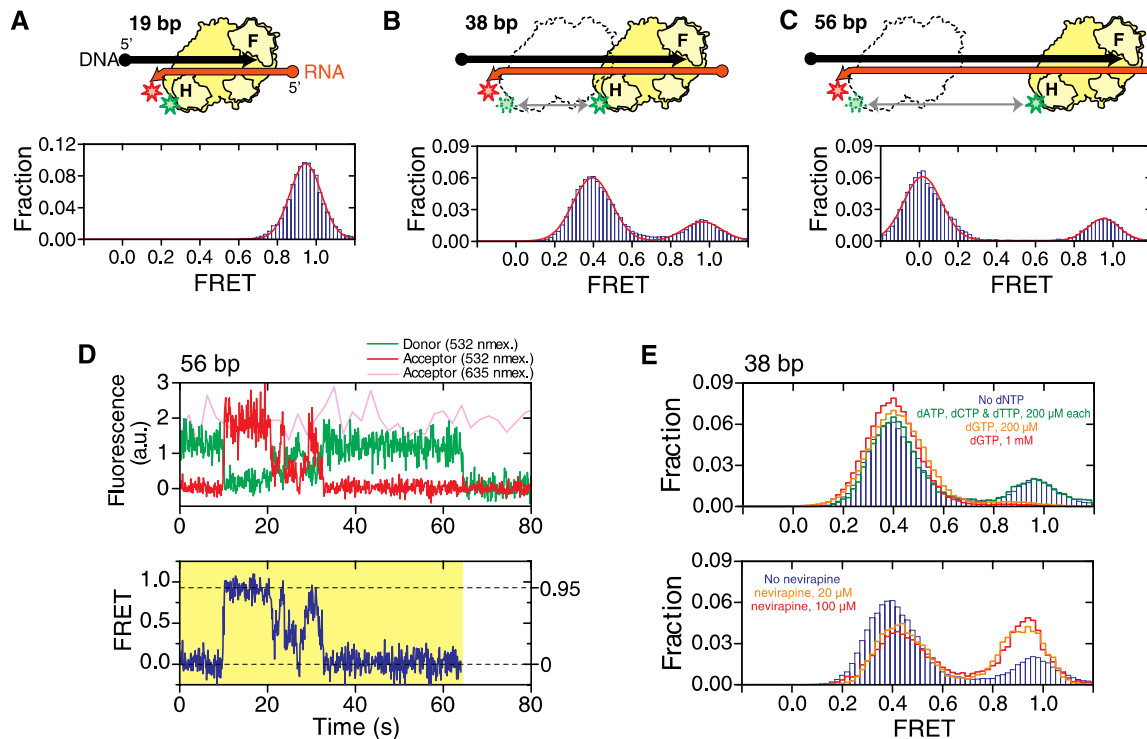
# Measuring Action Potential with FRET



# Measuring molecular mechanics by using FRET

## Slide into Action: Dynamic Shuttling of HIV Reverse Transcriptase on Nucleic Acid Substrates

Shixin Liu,<sup>1</sup> Elio A. Abbondanzieri,<sup>1</sup> Jason W. Rausch,<sup>4</sup> Stuart F. J. Le Grice,<sup>4</sup> Xiaowei Zhuang<sup>1,2,3\*</sup>



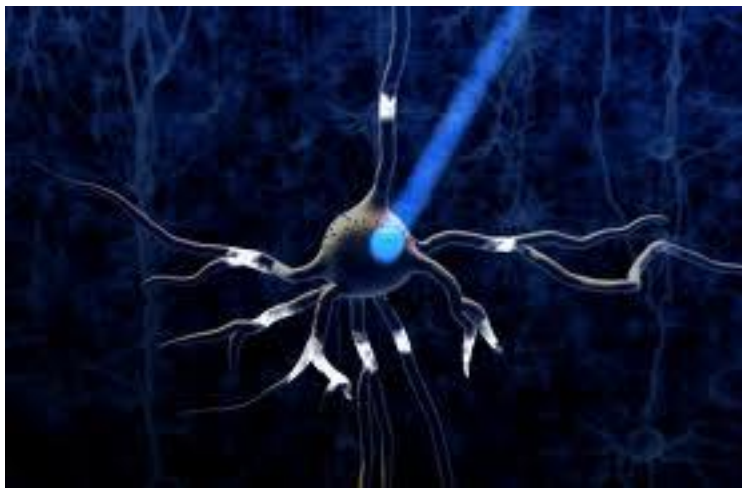
# What tools do we need to advance biology in the future?

## Computational Tools



1. Advance programs

## Biological Tools



1. New biological fluorescent reporters
2. Optigenetic switch
3. Optical and imaging methods

# Biological tool (optogenetics) to understand brain

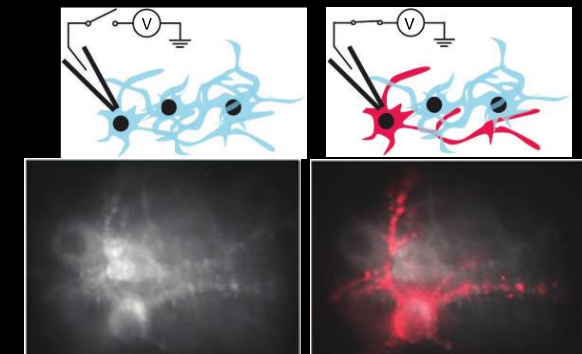
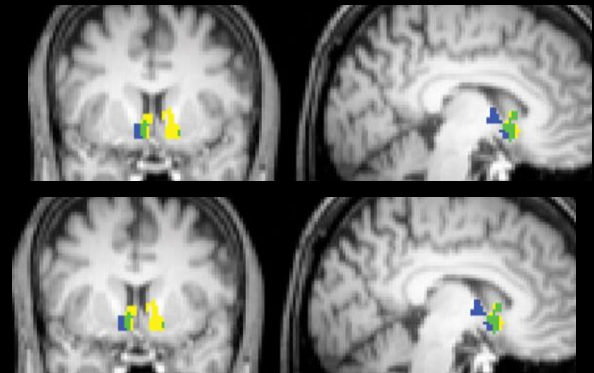
- Optical control of neuronal activity

Prof. Karl Diesseroth-Stanford

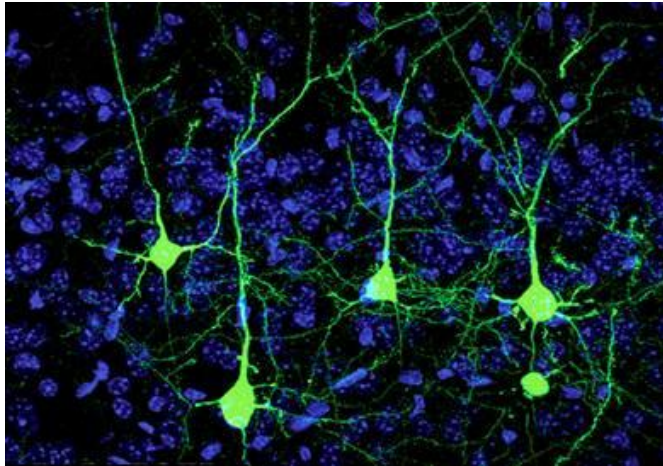
Prof. Ed Boyden-MIT

- Optical probes for membrane environment

Prof. Adam Cohen-Harvard



# New optical tools and probes are needed to increase spatiotemporal resolution.



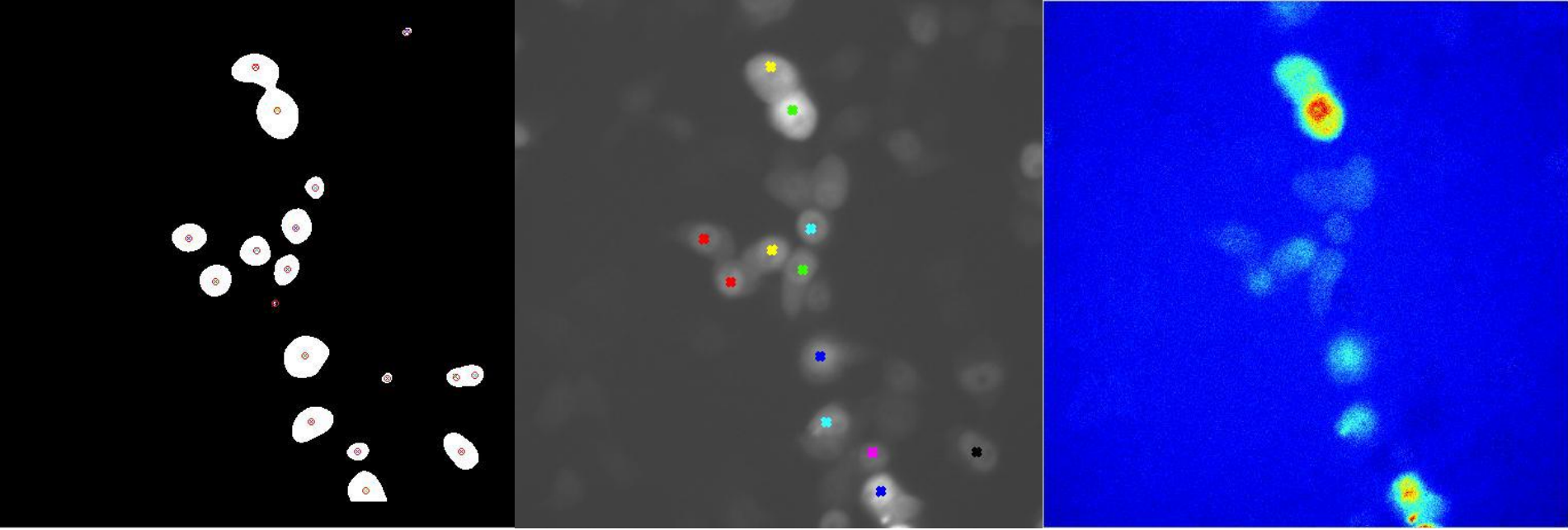
Static Imaging of cell function  
Low spatiotemporal resolution



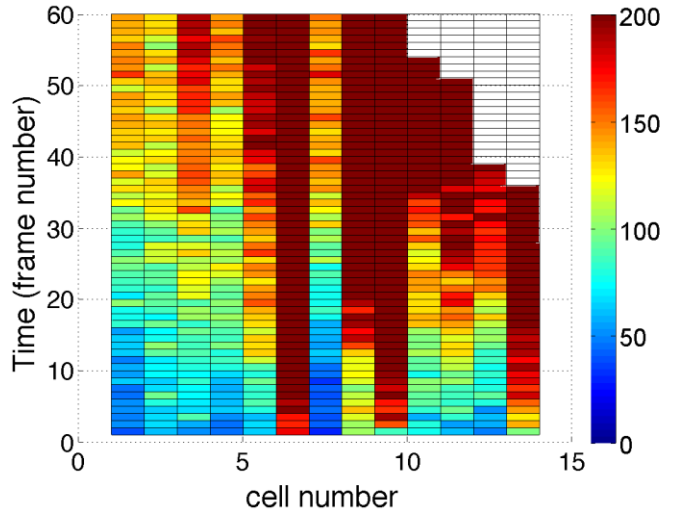
Dynamic Imaging of cellular function  
High spatiotemporal resolution



# Optogenetic tools to understand Cancer



We need advance computational tools and biological probes to solve problems in cancer



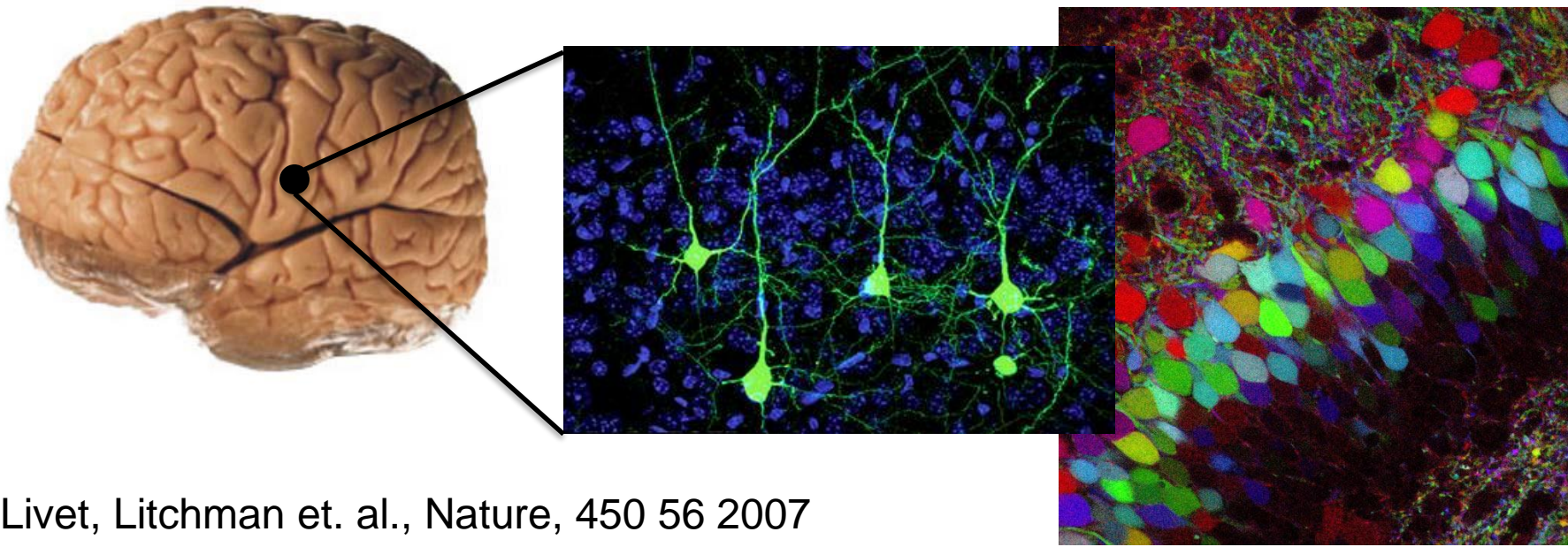
Data and Video Courtesy - Z. Kaya, N. Lack, H. Bayraktar



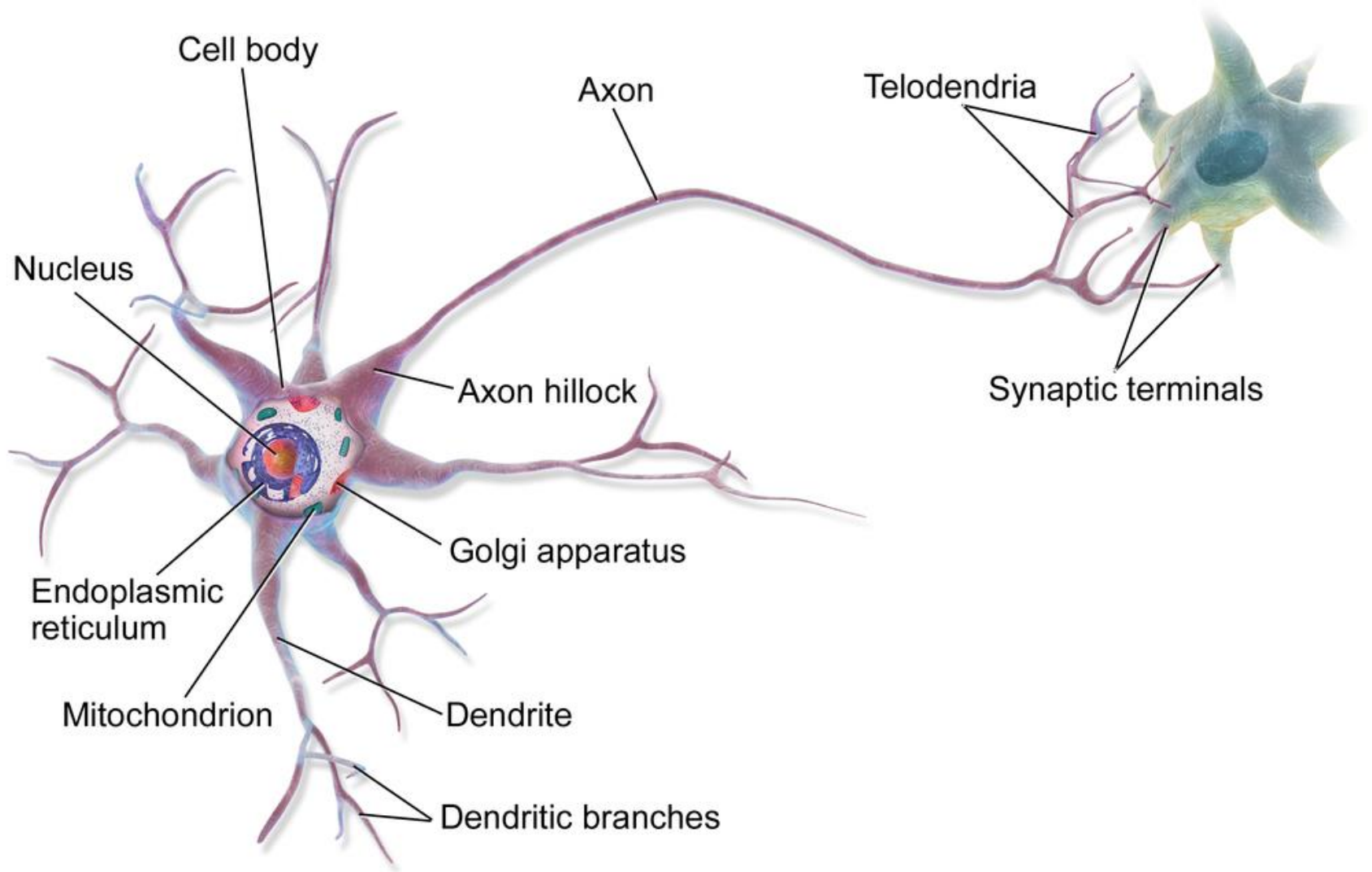
Advance Problems:

# Brain is a very complex organ

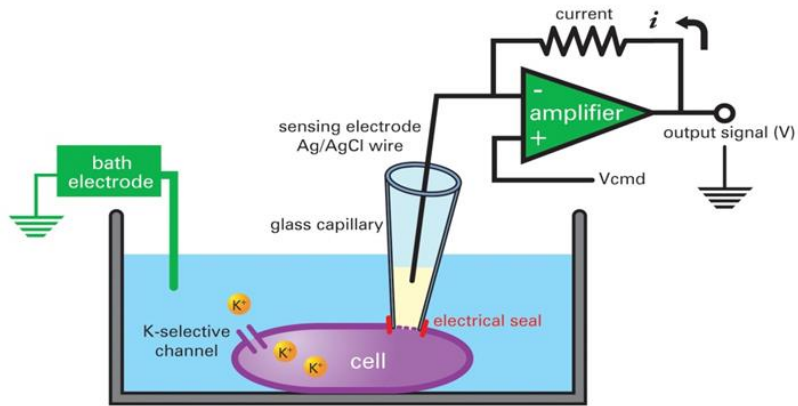
- Millions of neurons forms a highly complex architecture.
- Neurons are specialized to control certain function.
- A neuron subset are localized with other subsets so distinguish them are very difficult.



# Neuron

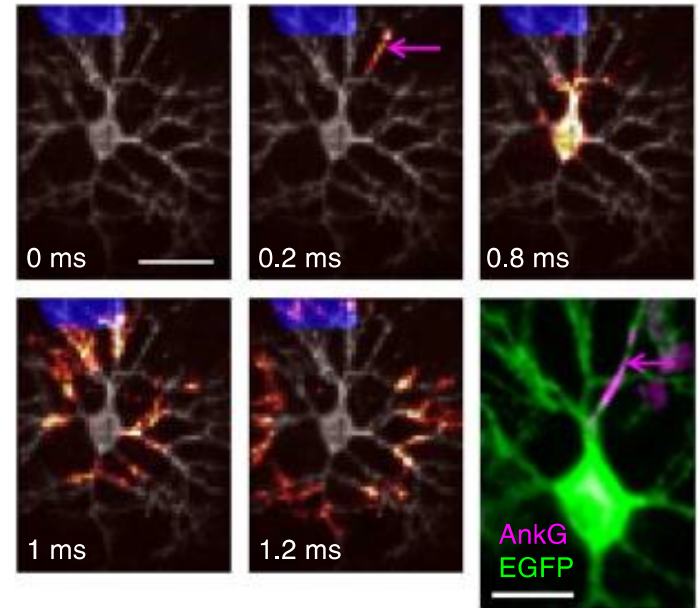
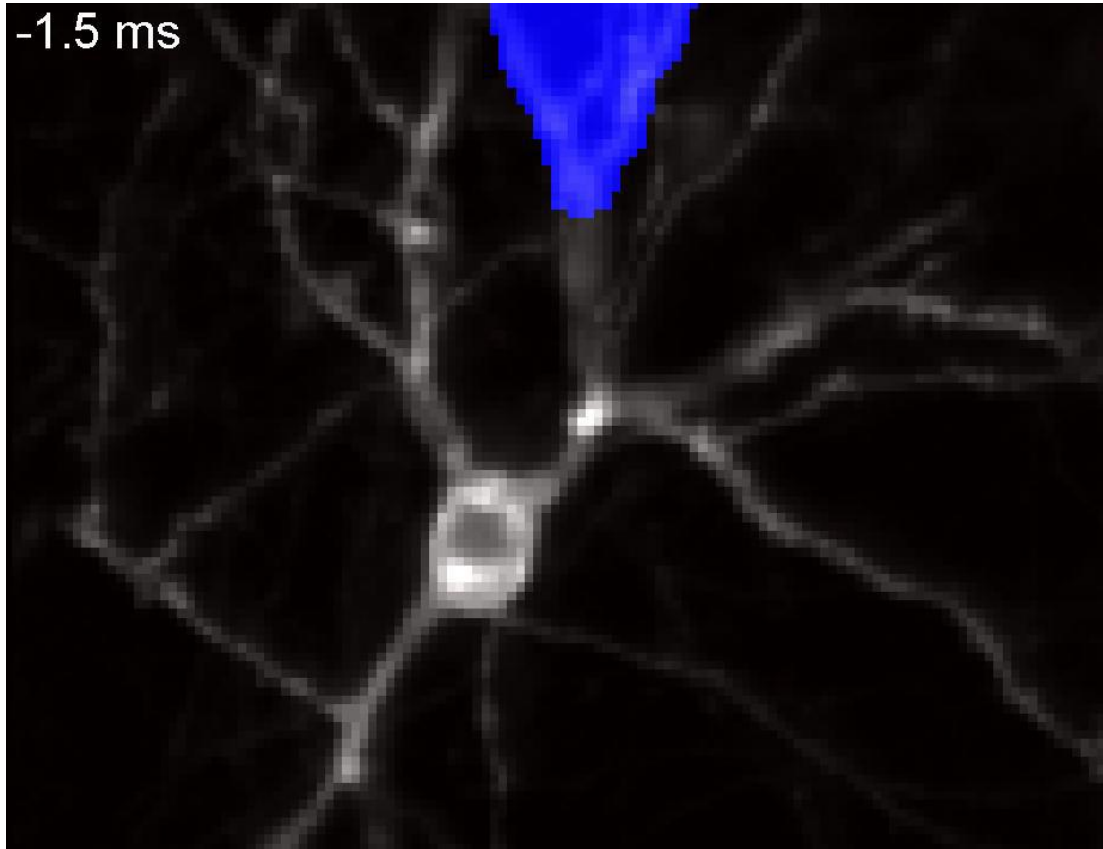


# Conventional Methods to measure membrane potential



80 years old conventional method

# All-optical electrophysiology in mammalian neurons using engineered microbial rhodopsins

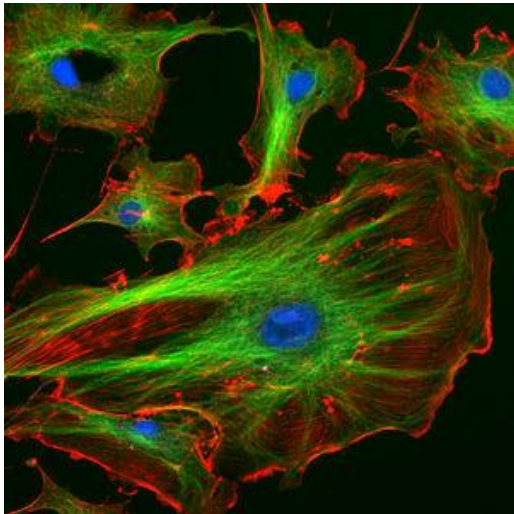


Video Courtesy Adam Cohen Harvard University  
Nature Methods , 2014

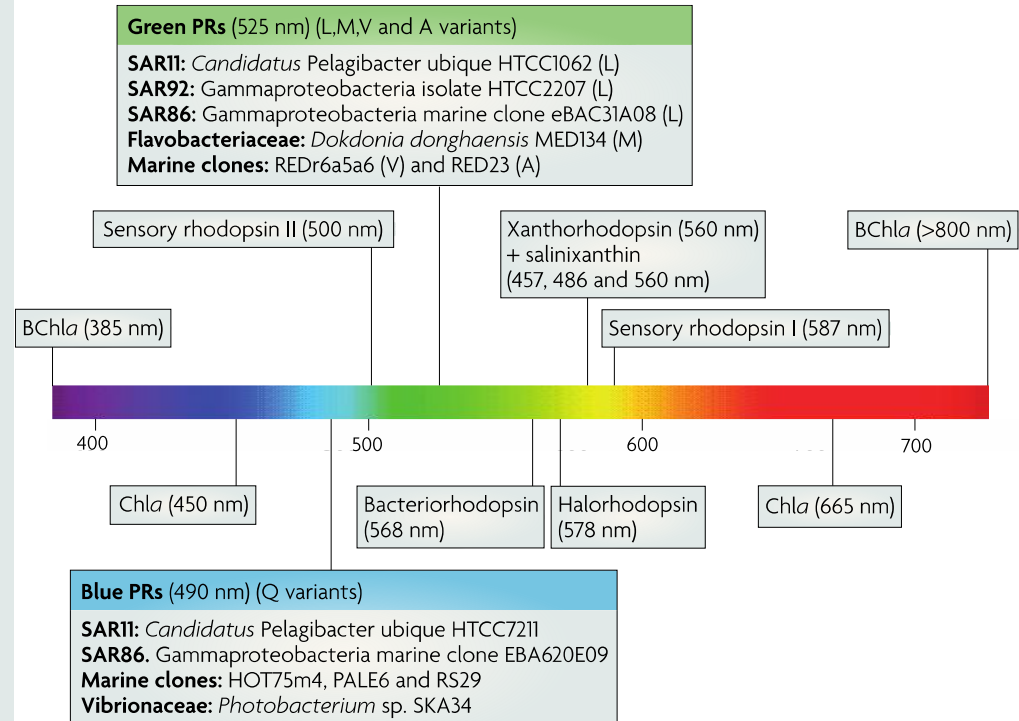
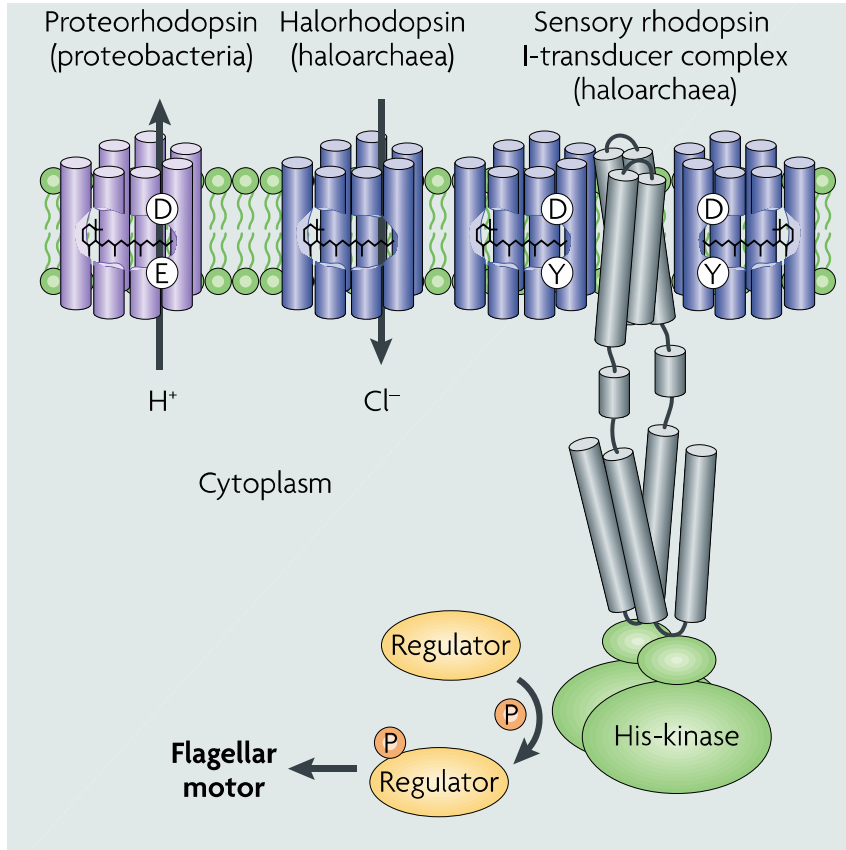


# Optic + genetic = Optogenetic

**Optogenetic** is the combination of genetic and optical methods to control and to monitor specific events in targeted cells of living tissue.



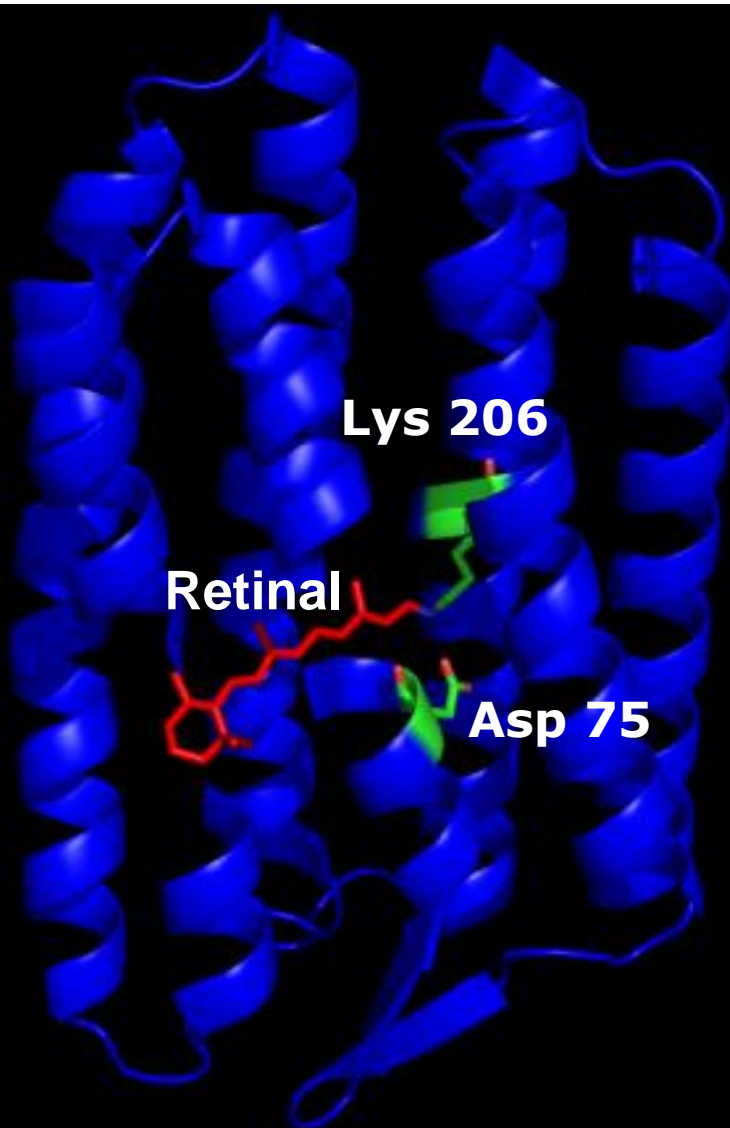
# Photoactive membrane proteins



# Channel rhodopsin

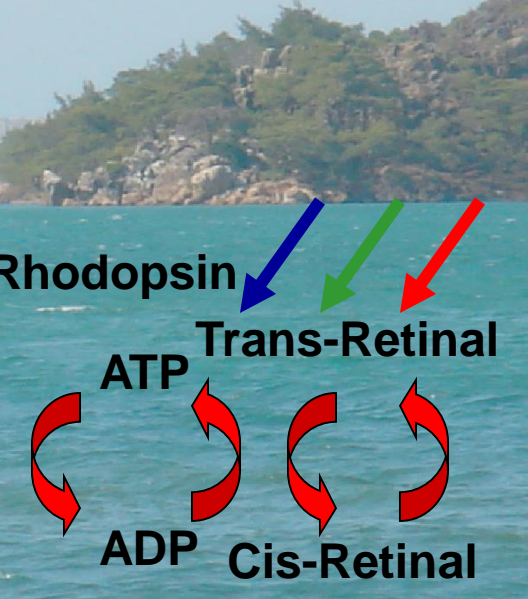
Light gated Ion  
channel  
It control phototaxis  
in algae.

ChR2 absorbs blue light with  
A max wavelength 490 nm

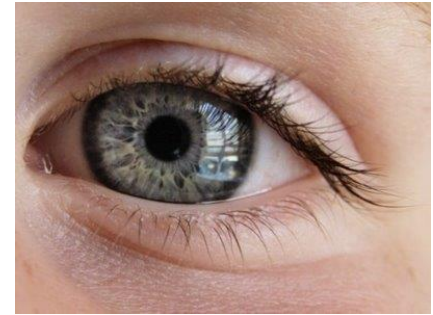


# Photoactive membrane proteins: energy harvesting, phototaxis, vision and optical control

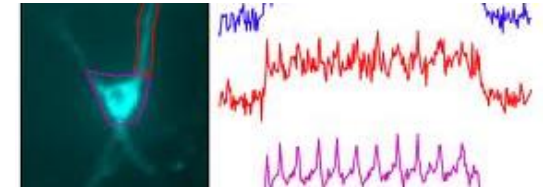
Energy harvesting



PHOTOACTIVE  
RHODOPSIN  
MEMBRANE  
PROTEIN



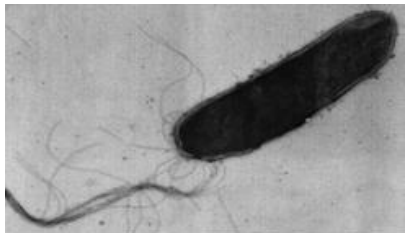
Vision



Optogenetic Applications

- Optical control of neuronal activity
- Optical probes for membrane potential

Phototaxis



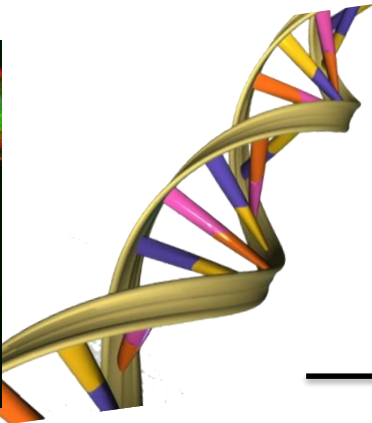
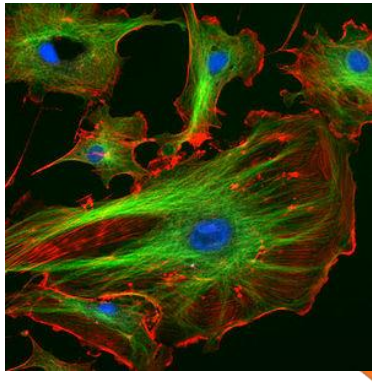
Halobacterium salinarum



The a photon is absorbed by the retinal. The *trans*-retinal complex change its conformation from trans to cis.

# The emerge of optogenetic methods

“**Optogenetics** is the combination of genetic and optical methods to control and to monitor specific events in targeted cells of living tissue.” Resource : Wikipedia

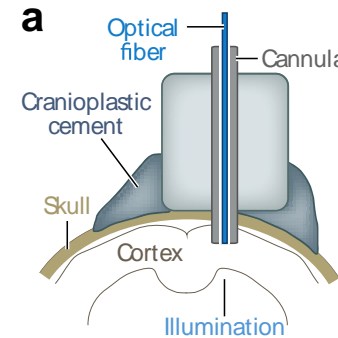
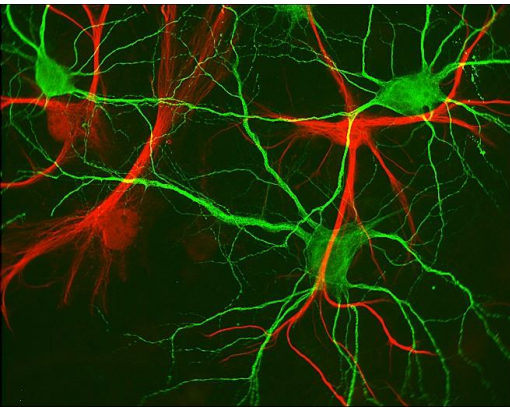


Dr. Karl Deisseroth Lab - Stanford



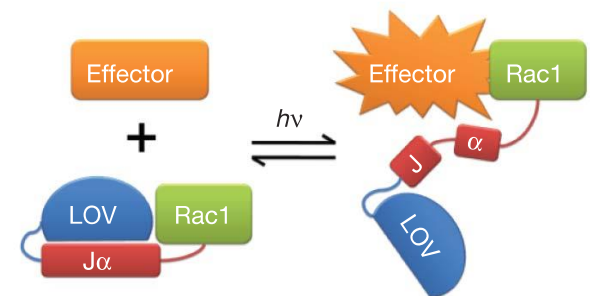
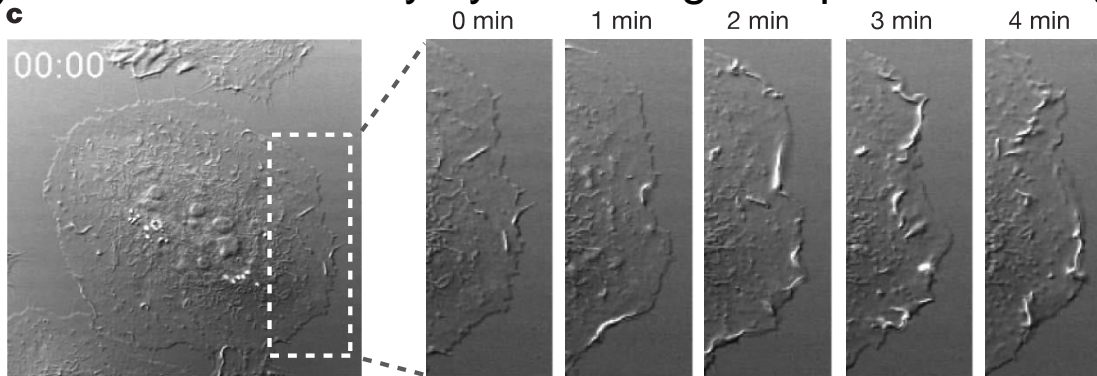
# 1. Optical control of cellular functions

A) Control of cell potential by activating Rhodopsin membrane proteins using lasers



Hippocampus : Controls movement

B) Control cell motility by activating Rac proteins using lasers



# How to modulate neurosignalling?

Small molecules can be used to affect the synapse activity

They are

Noepinephrine

Acetylcholine

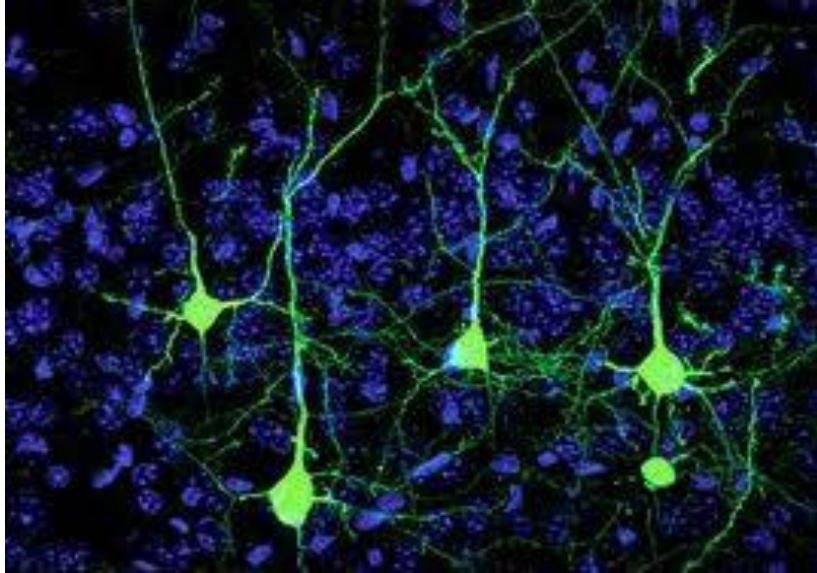
Dopamine

Serotonin

# Here is the problem:

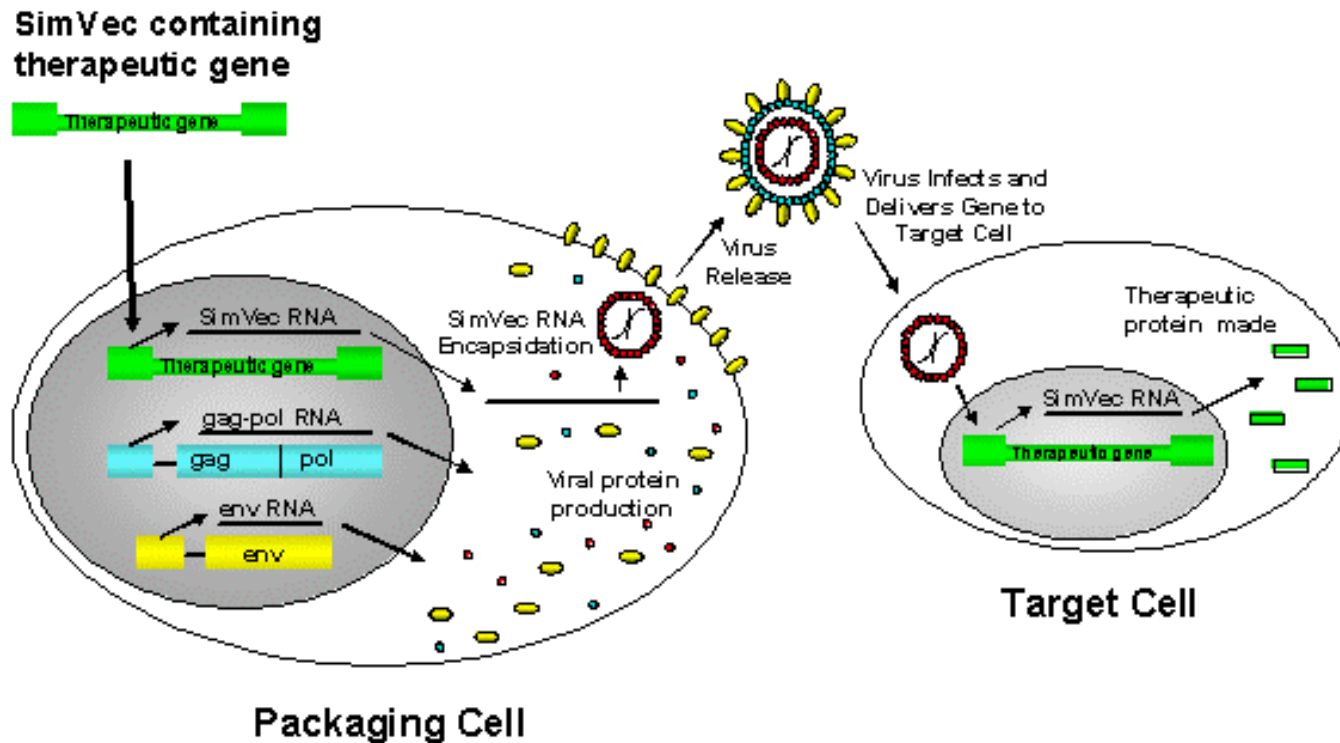
- Neurons system is composed by a large subset of specialized neurons.
- If a neurons needs to be studied, it has be modulated without modulating other things belongs to a different set of speciazed cells
- Small molecules will be localized in all cells because of the diffusion and affect all neurons.
  
- The question is can we modulate individual neurons? If how?
  - a) Optic methods
  - b) Genetic methods

# Lasers to modulate neurons

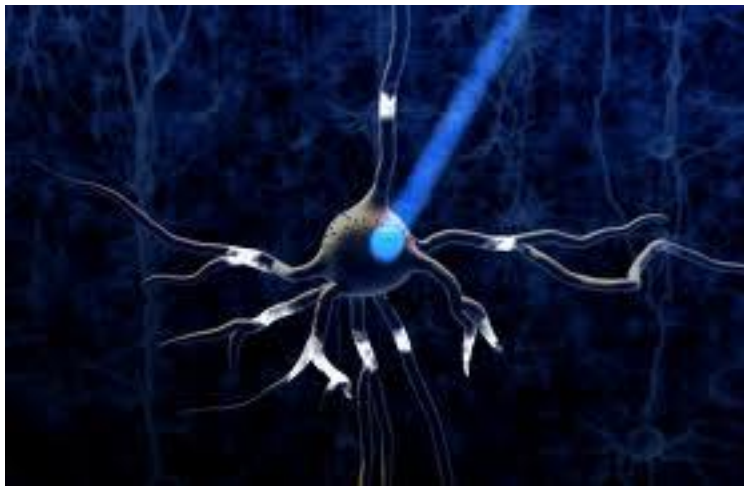
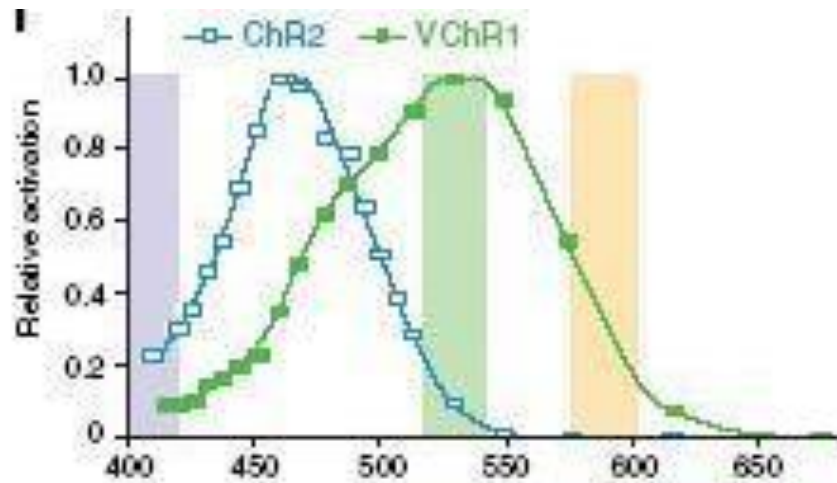


# Lentivirus for transferring genes

293Ft cells can be used for producing lentivirus.

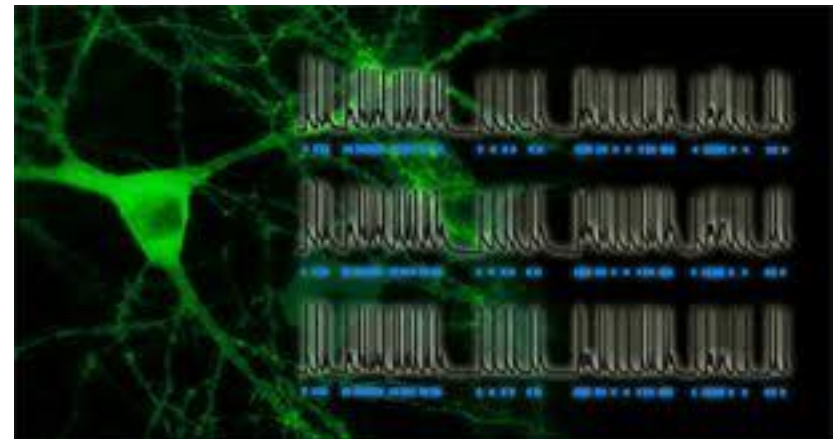
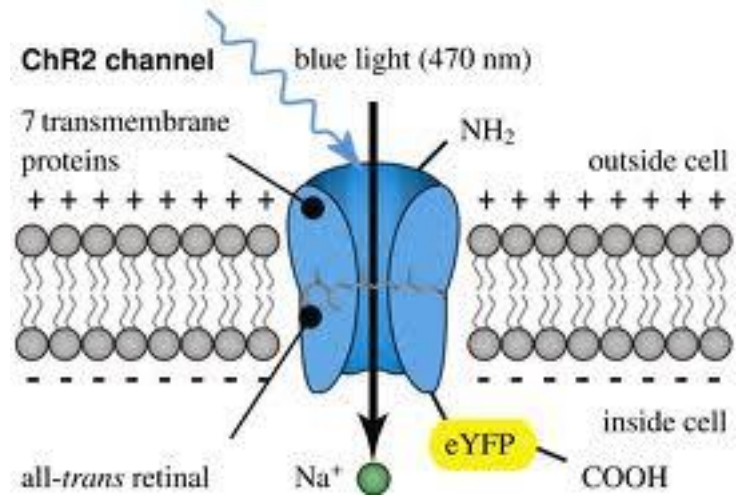
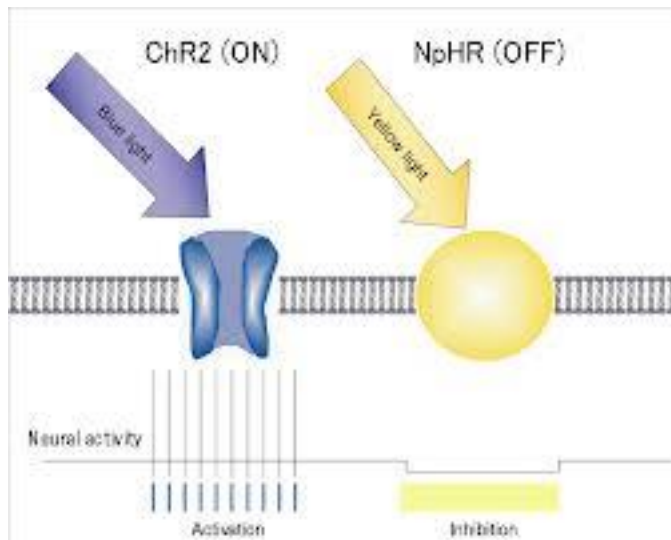


# Channel Rhodopsin





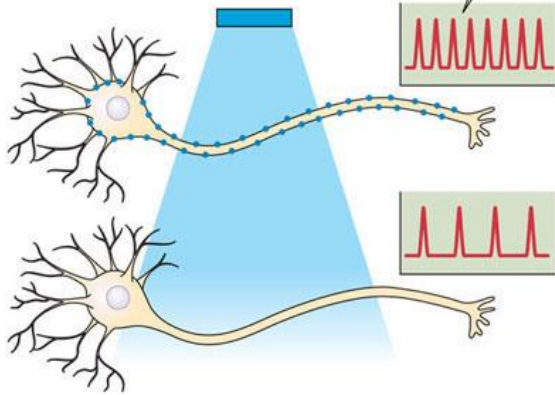
# How Channel Rhodopsin Activates Neurons?



# Rate of firing can be modulated

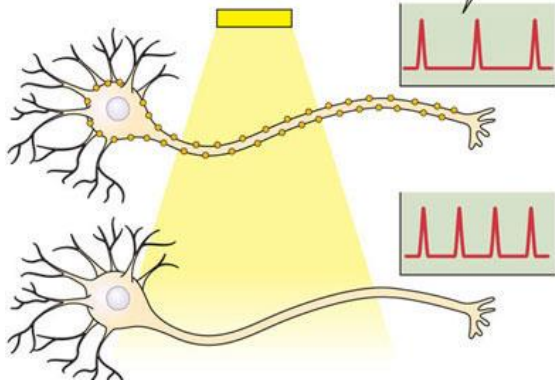
Optogenetic excitation

Shining blue light on a neuron expressing channelrhodopsin increases its firing rate.



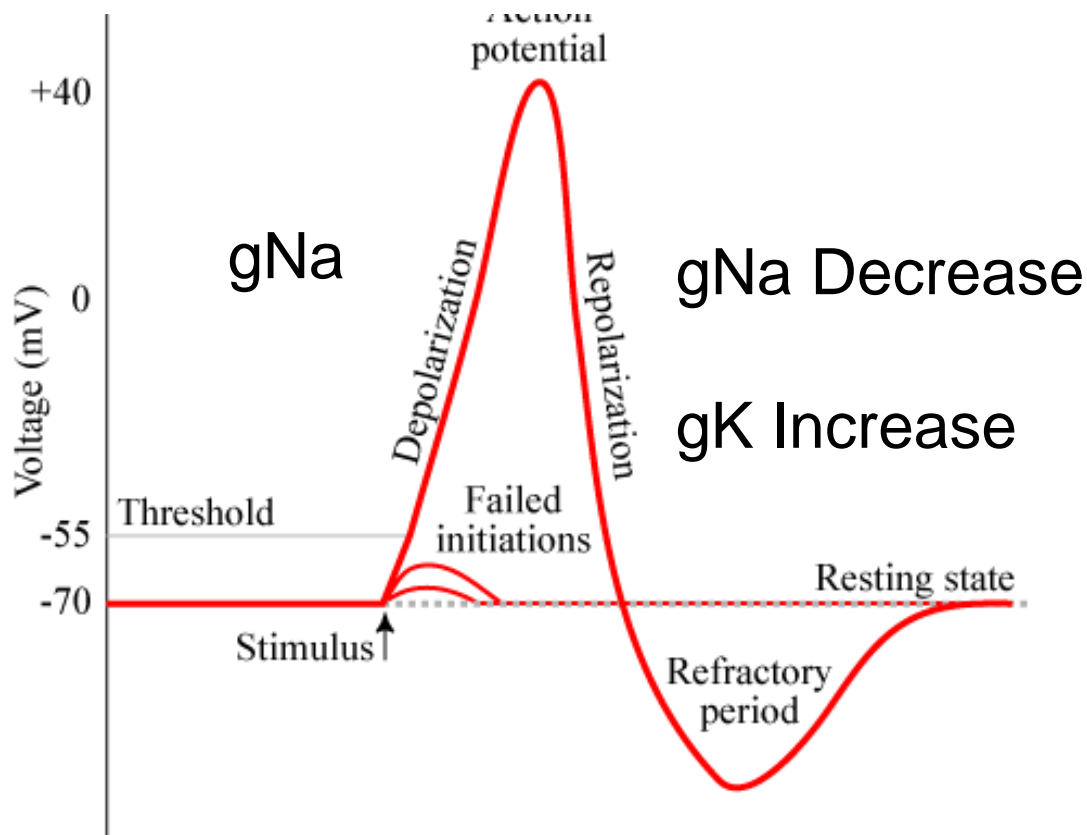
Optogenetic inhibition

Shining yellow light on a neuron expressing halorhodopsin decreases its firing rate.



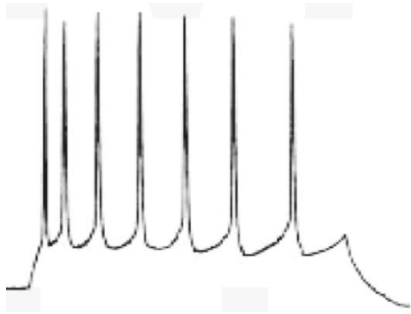
Frequency of spiking in a given time interval. When the brain for your hand holding a paper, the motor neurons controls the muscles by increasing firing rate.

# Action Potential

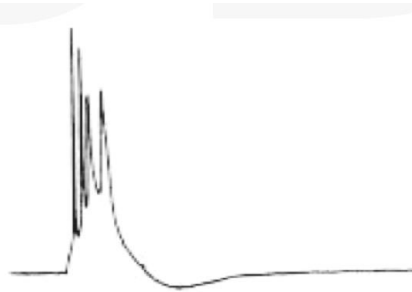


# Ion channel diversity leads to a neuronal diversity

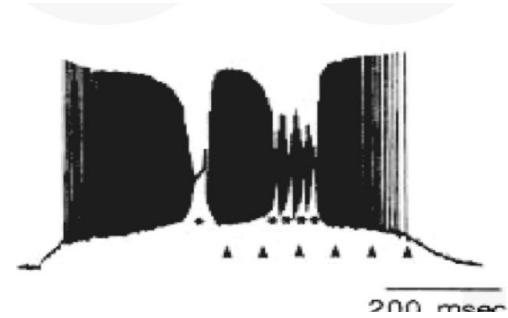
Regular Firing



Burst Firing

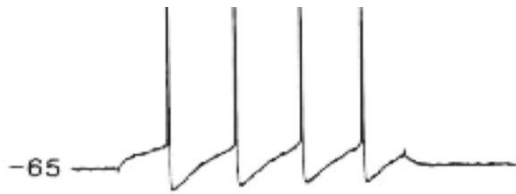


Cerebellar Purkinje Cell

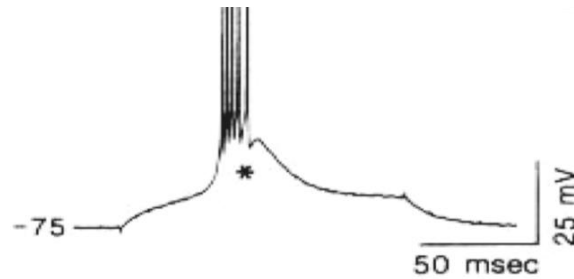


Cornical Pyramidal Cell

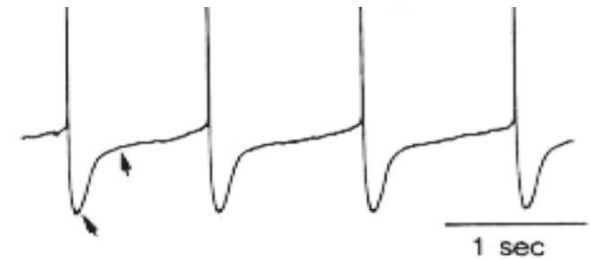
Medial Habenular Cell



Transfer mode



Burst mode



Chemical signaling is OK for short distances

Time constant for one dimensional diffusion:

$$\tau = \frac{x^2}{2D}$$

For a small molecule  
( $D \sim 10^{-5} \text{ cm}^2 / \text{s}$ )

where,  $x$  is the displacement in one dimension

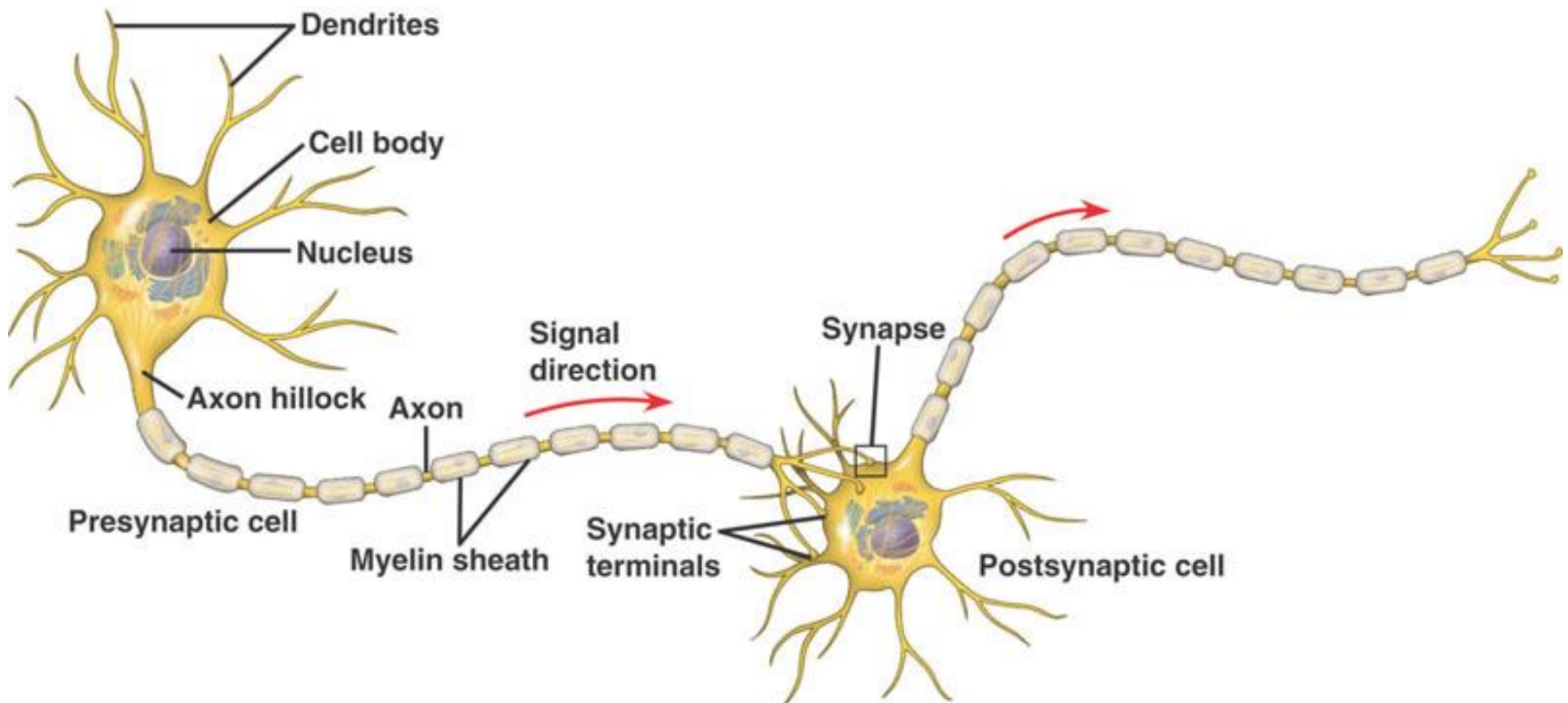
$D$  is the diffusion coefficient

$x$	$t$
1 $\mu\text{m}$	0.5 ms
100 $\mu\text{m}$	5 s
1 cm	13.9 hrs
1 m	16 years

Motor proteins move at only 10  $\mu\text{m/s}$ , or 27.8 hrs for 1m.  
In contrast, for electrical signals it takes 10ms to travel 1m.  
Electrical signals tend to be faster over long distances.

On dendrites, the analog signal is received through synapse

Signal is transmitted by action potential inside the neuron and exchange it across synapses



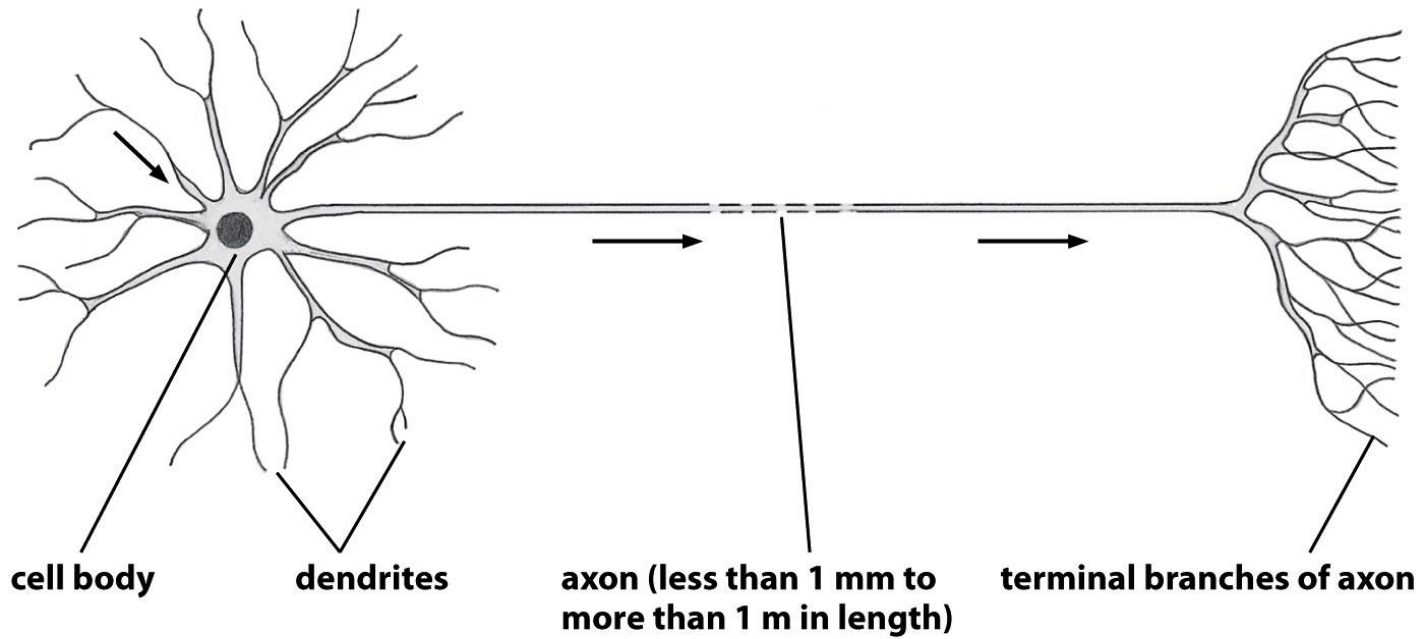
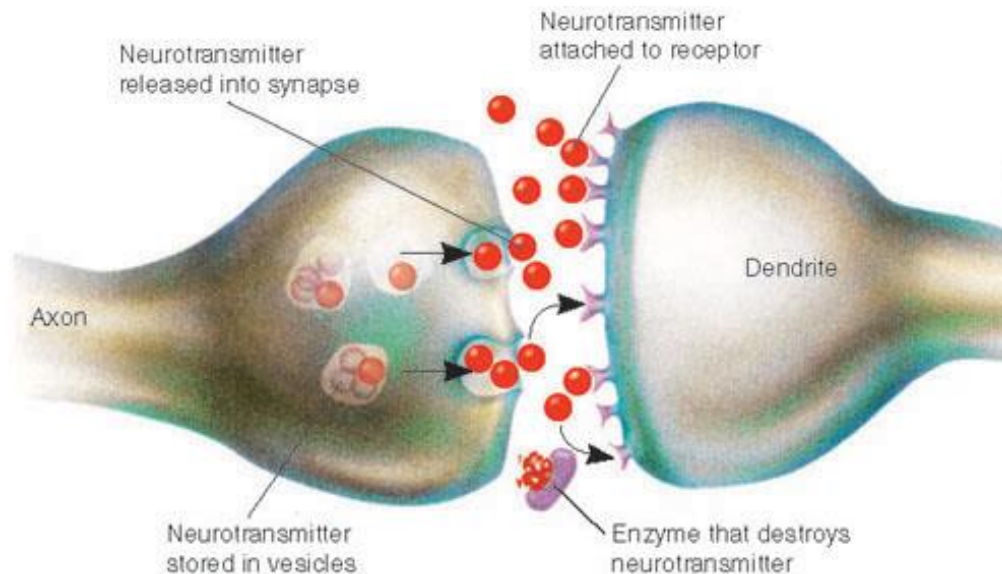


Figure 17.1 Physical Biology of the Cell (© Garland Science 2009)



The Axon,

Generally one per neuron

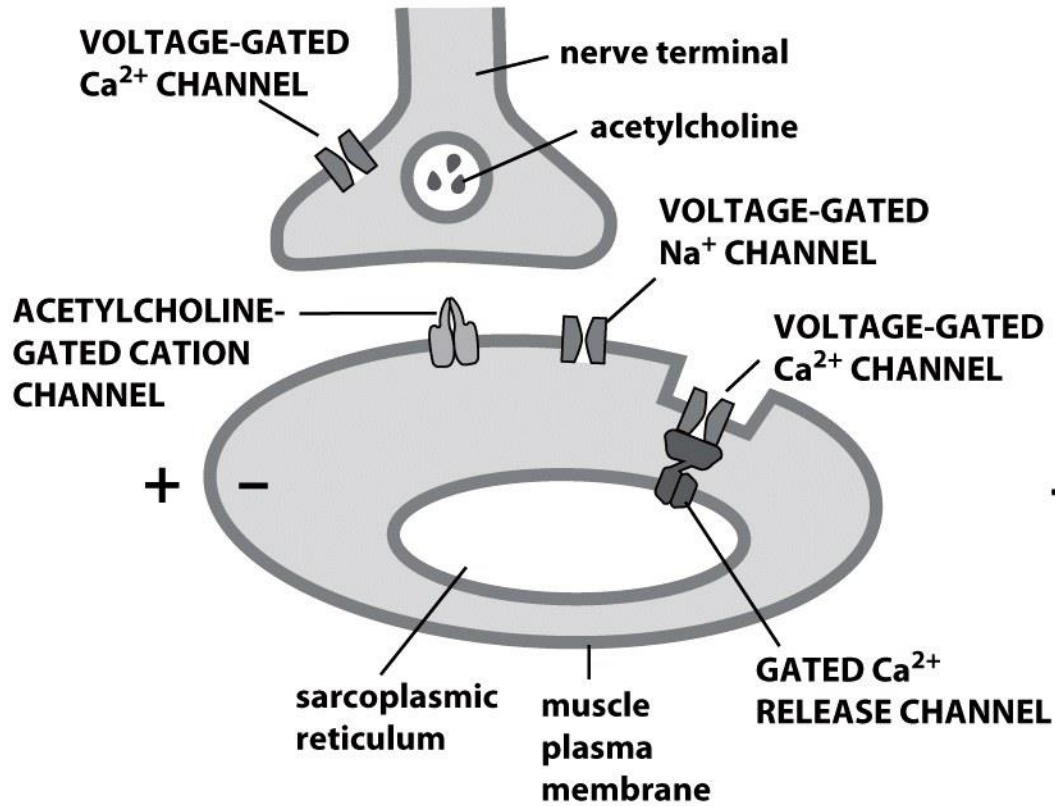
Many microtubules and neurofilaments are presents

A few micron to meter length

Transmit signal from some of the cell to the neuron  
synapses



## RESTING NEUROMUSCULAR JUNCTION



## ACTIVATED NEUROMUSCULAR JUNCTION

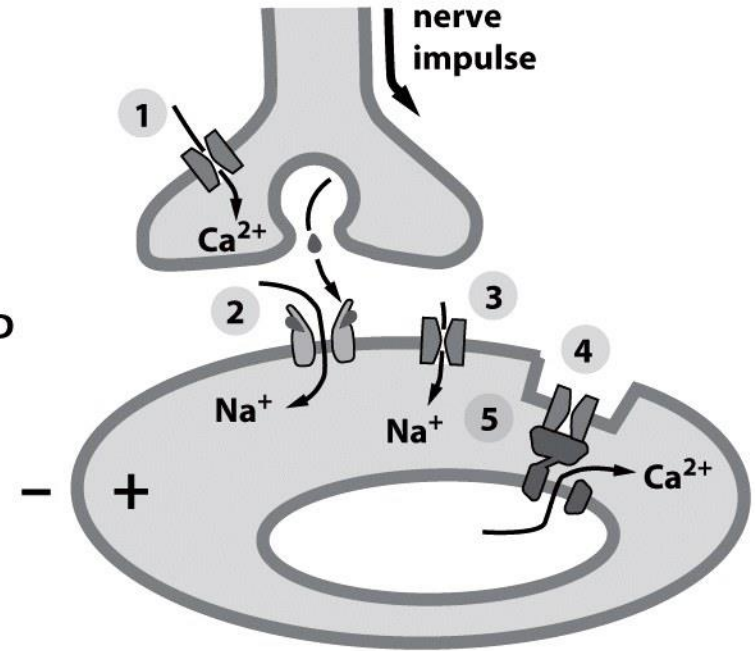


Figure 17.9 Physical Biology of the Cell (© Garland Science 2009)

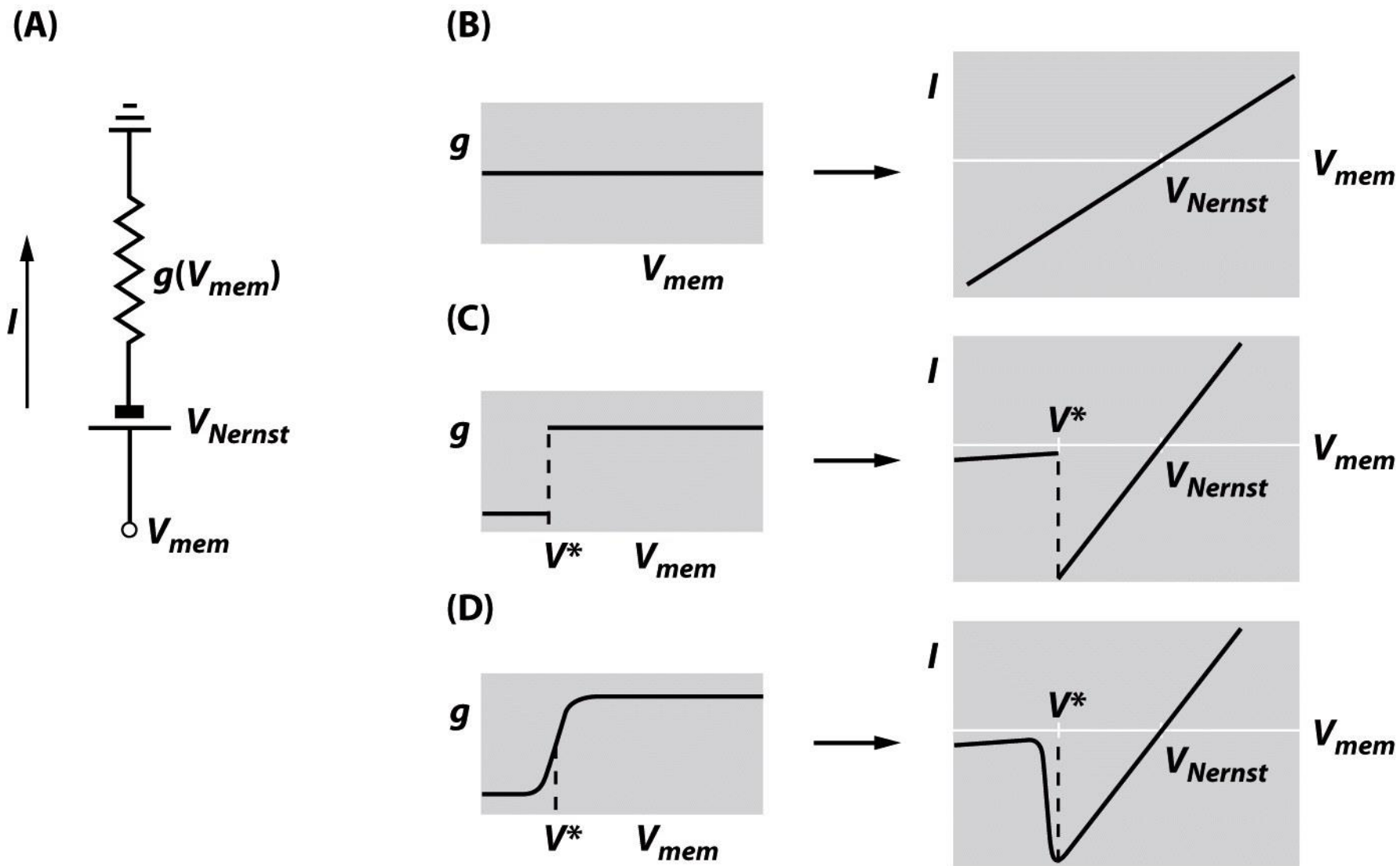
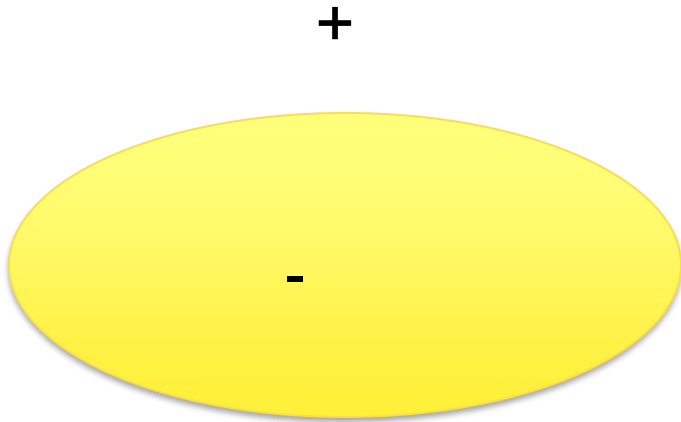


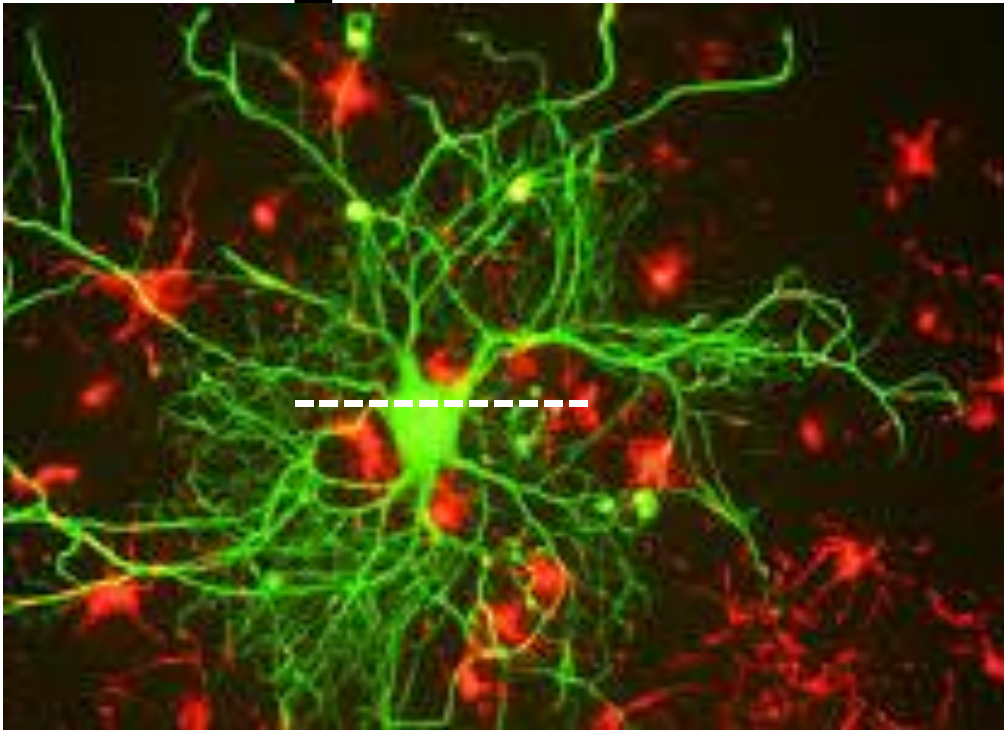
Figure 17.14 Physical Biology of the Cell (© Garland Science 2009)

Membrane potential of  $-70\text{ mV}$  means that the inside of the cell is negative compared to the outside (the outside of the cell is **always** of the opposite charge of what is inside).

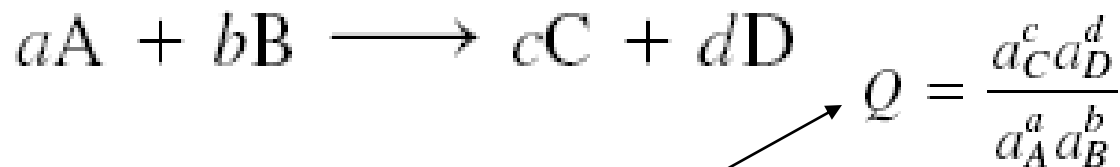


Membrane potential of -70 mV means that the inside of the cell is less negative compared to the

+++++



# The Nernst Equation and Concentration Dependence of the Emf



$$\Delta G = \Delta G^\circ + RT \ln Q$$

$$\Delta G = -nF\mathcal{E}$$

$$\Delta G^\circ = -nF\mathcal{E}^\circ$$

$$-nF\mathcal{E} = -nF\mathcal{E}^\circ + RT \ln Q$$

*Nernst equation*

$$\mathcal{E} = \mathcal{E}^\circ - \frac{RT}{nF} \ln Q$$

At 25°C (298.15 K),  $\mathcal{E}(25^\circ\text{C}) = \mathcal{E}^\circ - \frac{0.025693 \text{ V}}{n} \ln Q$

At equilibrium. no net transfer of electrons, so  $\mathcal{E}_{\text{cell}}^\circ = 0$ ,  $Q = K$ .

$$\mathcal{E}_{\text{cell}}^\circ = \frac{RT}{nF} \ln K$$

# Nerst Equation is used to calculate the equilibrium potential

There is a high negative charge outside of the cell

155 mM inside while the outside concentration of K is typically about 4 mM.

Remember that negative charges attract positive ones such as K<sup>+</sup> ions.

$$E = - \frac{RT}{nF} \ln \frac{K_i}{K_o}$$

$$E = - 59\text{mV} \ln \frac{K_i}{K_o}$$

$n$  is the elementary charges

$F$  is faraday constant

$T$  is temperature

$R$  is gas constant

$$\log(155/4)$$

$$E_K = -59$$

$$E_K = -98 \text{ mV}$$

If the membrane were permeable only to K,  $E$  would be -90 mV. Since the electrical force exactly balanced with the diffusion force when the membrane potential was -92 mV.

If the cell membrane is only permeable to Na ions. Then what is the membrane potential?

$$E_K = -59 \log \frac{Na_i}{Na_o}$$

If  $Na_{outside} = 145 \text{ mM}$   $Na_{inside} = 12 \text{ mM}$

Then,

$$E = -59 \text{ mV} \log(12/145)$$

$$E = +67 \text{ mV}$$

## Summary:

$$V_m = -90 \text{ to } -70 \text{ mV}$$

$$E_K = -98 \text{ mV}$$

$$E_{Na} = +67 \text{ mV}$$

Since the membrane is more permeable to K than Na, The  $V_m$  is more close to  $E_K$  than  $E_{Na}$



Ion species	Intracellular concentration (mM)	Extracellular concentration (mM)	Nernst potential (mV)
K <sup>+</sup>	155	4	-98
Na <sup>+</sup>	12	145	67
Ca <sup>2+</sup>	10 <sup>-4</sup>	1.5	130
Cl <sup>-</sup>	4	120	-90

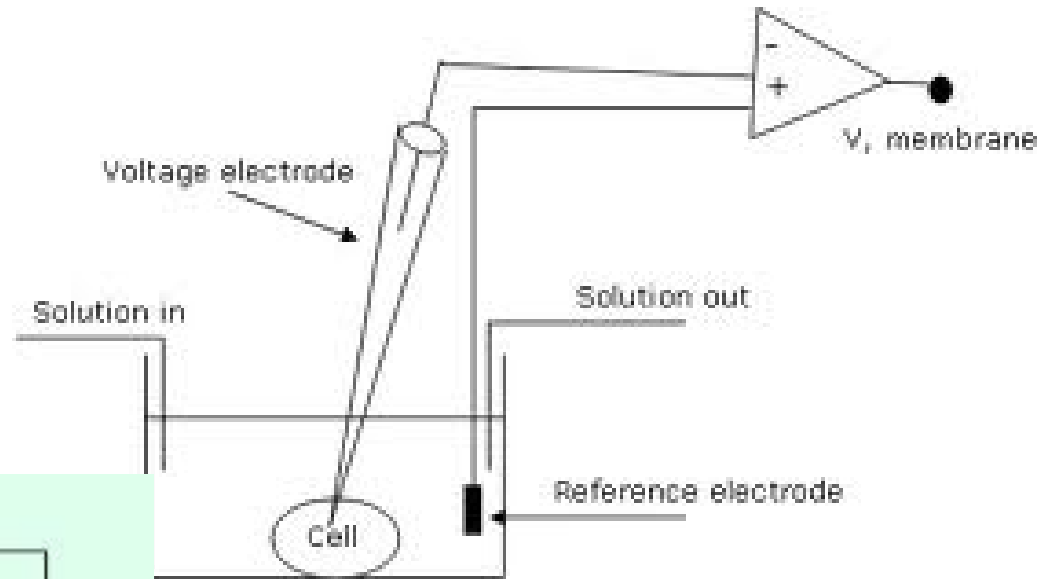
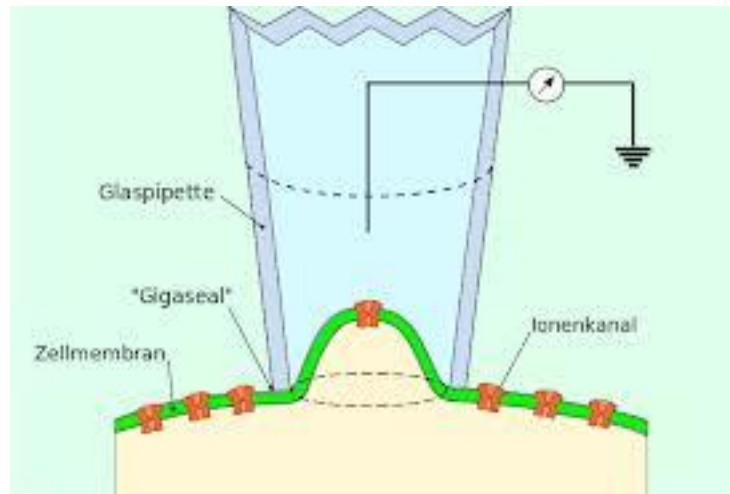
Table 17.1 Physical Biology of the Cell (© Garland Science 2009)



# How the neuron activity can be measured?

## Patch Clamp

Nobel Prize in Physiology & Medicine -1991



Extracellular

# Ion Channels

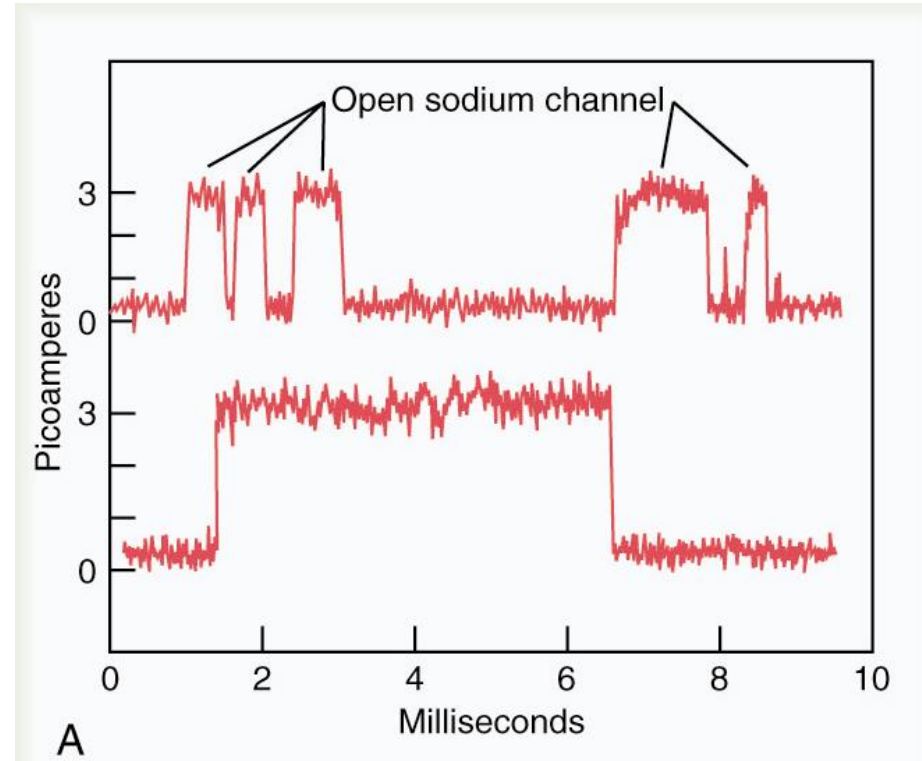
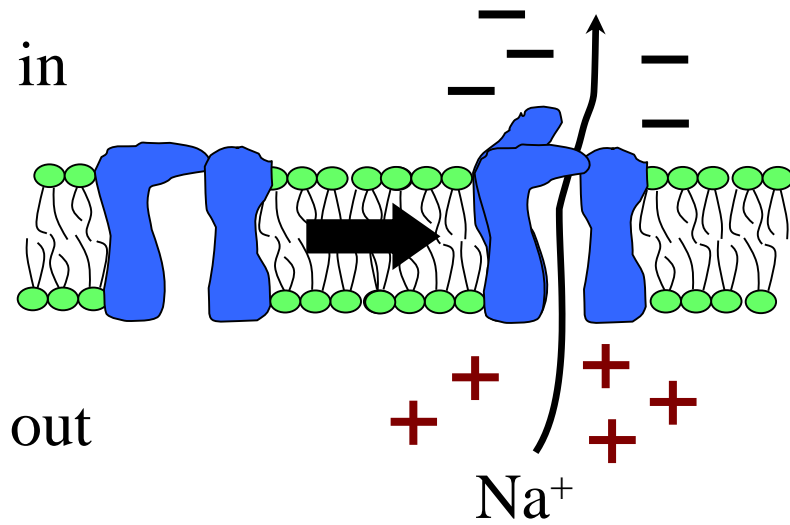
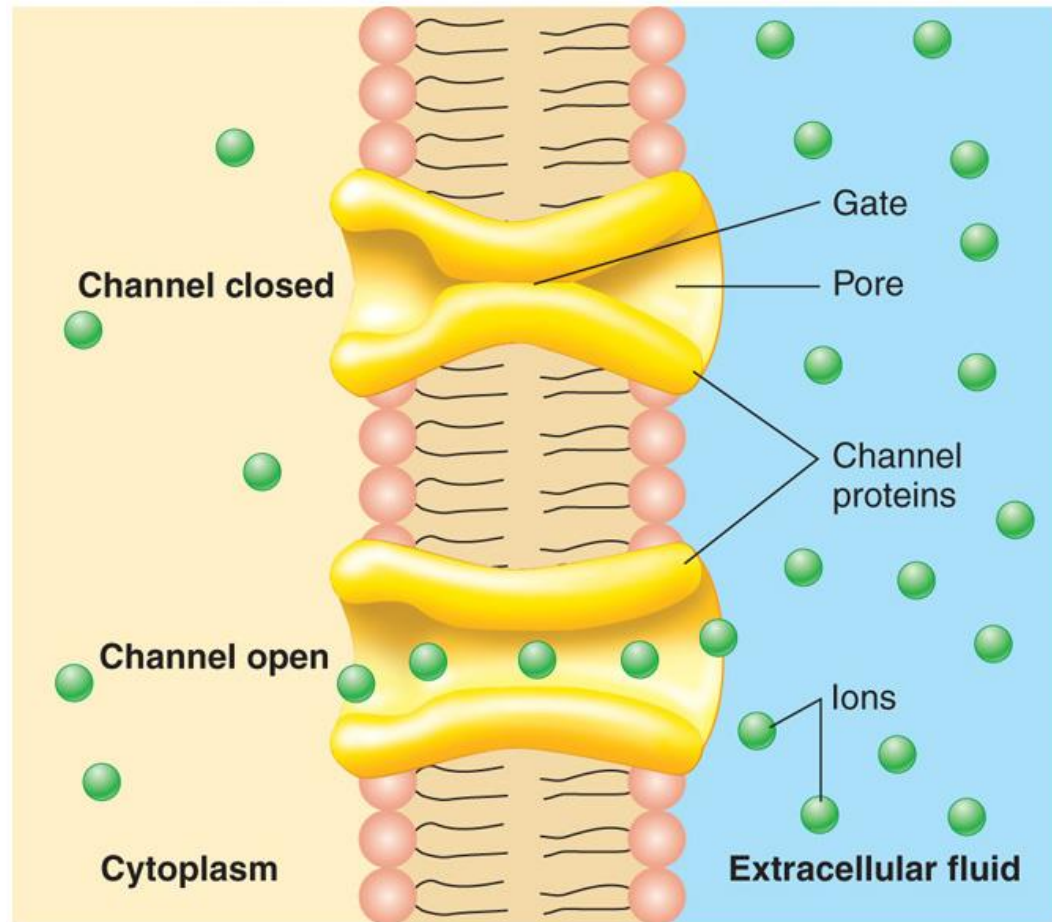


Figure 4-5; Guyton & Hall

# Diffusion continued

- **Cell membranes are impermeable to charged and most polar compounds**
  - Charged molecules must have an ion channel or transporter to move across membrane



(\*\* *across an artificial lipid bilayer*)

